

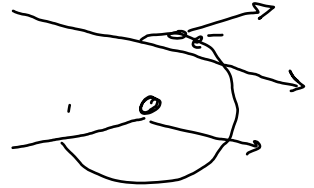
# Electromagnetic Interactions of Charged Particles w/ Matter

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- inelastic collisions with atomic electrons
- elastic scattering from nuclei

These processes give

- loss of incident particle energy
- change in particle trajectory



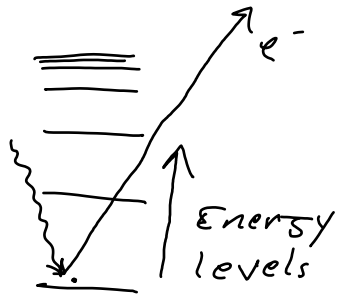
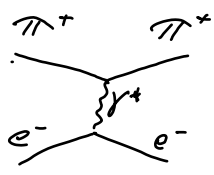
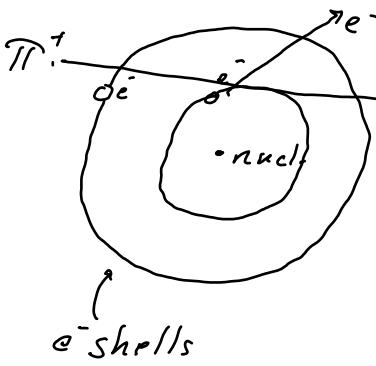
Additional reactions

- Cerenkov emission
- transition radiation
- bremsstrahlung
- scintillation

Aside from high energy electrons and positrons

- inelastic collisions dominate electromagnetic interactions

# Ionization and Inelastic Collisions



charged particle interacts with atomic electrons

ionization or excitation results

Only few eV ~~are~~ lost; but many interactions in material

Remember  $\sigma_{atom}$ !

Example: 10 MeV proton incident on copper plate

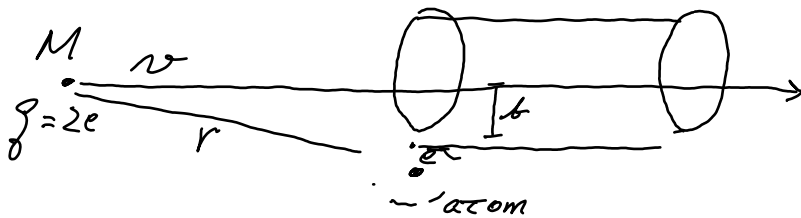
→ total energy loss in 1/4 mm

Some collisions provide sufficient energy to ionization electrons

→ cause ionization themselves

" $\delta$  rays"

# Classical Calculation of Energy Loss



Coulomb interaction  
with  $e^-$ , nucleus

## Assumptions

- $M \gg m_e$
- electron free and initially at rest
- electron only moves slightly during interaction
- $E$ -field remains as in initial position
- incident particle undeviated

Momentum gained by electron

$$\begin{aligned} \Delta p &= \int F dt = e \int E_{\perp} dt = e \int E_{\perp} \frac{dt}{dx} dx \\ &= \frac{e}{v} \int E_{\perp} dx \end{aligned}$$

(only care about  $E_{\perp}$  due to symmetry.)

## Momentum Transfer

Use Gauss's Law for the integral

$$\int \vec{E}_\perp \cdot d\vec{A} = 4\pi z e$$

$$= \int E_\perp 2\pi b dx$$

$$\int E_\perp dx = 2ze/b$$

This gives

$$\Delta p = \frac{e}{v} \int E_\perp dx = \frac{2ze^2}{bv}$$

$$\Delta E = \frac{\Delta p^2}{2m_e} = \frac{2ze^4}{m_e b^2 v^2} \propto \frac{1}{b^2 v^2}$$

$\therefore$  most of energy transfer due to close collisions

Where is energy transfer happening?

atomic electrons:  $m = m_e$ ;  $e^2 \rightarrow e^2$

nuclei:  $m = A m_p$ ;  $e^2 \rightarrow Ze^2$

$$\frac{\Delta E(e)}{\Delta E(\text{nucl.})} = \frac{2ze^4}{m_e b^2 v^2} \cdot \frac{b^2 v^2 A m_p}{2z^2 Z^2 e^4} \sim \frac{2z m_p}{Z^2 m_e}$$

$$= 2m_p / Z m_e \gg 1$$

$\therefore$  atomic electrons mostly where happening

# Energy Loss per Unit Length

If  $N_e$  is density of electrons

$$d\mathcal{E}(b) = \Delta\mathcal{E} N_e dV = \frac{22^2 e^4}{m_e b^2 v^2} N_e (2\pi b db dx)$$

energy lost  
to all electrons  
between  $b$  and  
 $b+db$  in  $dx$

$$= \frac{4\pi 2^2 e^4}{m_e v^2} N_e \frac{db}{b} dx$$

Need to integrate over  $b$  to get

$$\frac{d\mathcal{E}}{dx} = \frac{4\pi 2^2 e^4}{m_e v^2} N_e \int_{b_{min}}^{b_{max}} \frac{db}{b}$$

Problem: 1) a large  $b$  takes a long time:  $e^-$  moves

2) as  $b \rightarrow 0$ ,  $\Delta p \rightarrow \infty$

$$\therefore \frac{d\mathcal{E}}{dx} = \frac{4\pi 2^2 e^4}{m_e v^2} N_e \ln \frac{b_{max}}{b_{min}}$$

Minimum impact parameter:

→ maximum  $\Delta E$  when a head-on collision

$$\Delta E_{\max} = \frac{(2p)^2}{2m} = 2\gamma^2 m_e v^2$$

$$\text{also} = \frac{2z^2 e^4}{m_e v^2 b_{\min}^2}$$

$$\therefore \underline{b_{\min} = 2e^2 / \gamma m_e v^2}$$

Maximum impact parameter

$e^-$  has orbital frequency,  $\nu$

period  $T = 1/\nu$

eg  
O

Interaction with passing particle short compared to  $T$

$$\tau/\gamma = (b/v)/\gamma = b/\gamma v \leq T = 1/\nu$$

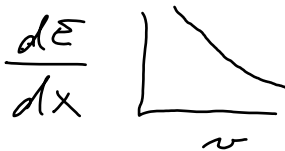
$$\therefore \underline{b_{\max} = \gamma v / \nu} \rightarrow \text{typical (on average) frequency}$$

Substitution gives:

$$\frac{dE}{dx} = \frac{4\pi z^2 e^4 N_e}{m_e v^2} \ln \left( \frac{\gamma^3 m_e v^3}{2e^2 \nu} \right)$$

$\frac{1}{v^2}$  @ low energy

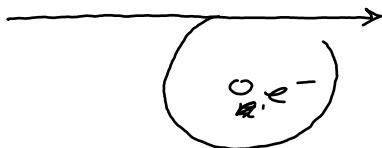
$\ln \gamma^2$  @ high energy



Ex: 21 Classical Electromagnetic  
Cross Section

(30)

When a particle travels within 'b' of atom



want the cross section as a function of Energy transfer  $W$

Differential cross-section in area of rings

$$d\sigma(b) = \frac{d\sigma(b)}{db} db = 2\pi b db$$

Using  $W \Delta E = 2z^2 Z^2 e^4 / b^2 v^2 m$ ,

Then:  $b = \sqrt{\frac{2z^2 Z^2 e^4}{W \Delta E v^2 m}}$ ;  $db = -\frac{1}{2} \sqrt{\frac{2z^2 Z^2 e^4}{v^2 m}} \frac{1}{W^2} dW$

This gives us

$$d\sigma(b) = 2\pi b db = \frac{2\pi z^2 Z^2 e^4}{v^2 m} \frac{dW}{W^2} = d\sigma(W)$$

$$\boxed{\frac{d\sigma}{dW} = \frac{2\pi z^2 Z^2 e^4}{v^2 m} \frac{1}{W^2}}$$

cross section of particle with energy  $E = E(v)$  to lose energy  $W$  in a collision with a free electron

# Quantum treatment of Energy Loss

(37)

Must account for:

- particle energy transfers to atomic electrons only in discrete amounts
- particles have a wave nature  
i.e. for small 'b', point-like particle is not valid

We think in terms of small (large) momentum transfers, not large (small) impact parameters

→ also Dirac wave-functions with Coulomb potential coupling to  $\gamma$  field

To get the total energy loss, need to integrate over all excitation energies weighted by cross section

$$\frac{dE}{dx} = n_a \int E d\sigma$$



In close collisions

(32)

$$\left. \frac{dE}{dx} \right|_{W > \eta} = n_e \int_{\eta}^{W_{\max}} W \frac{d\sigma}{dW} dW$$

$\eta$  = an energy cutoff limiting energy transfers  $\rightarrow$  50 keV

$W_{\max}$  = maximum energy transfer

For the differential cross section  $\frac{d\sigma}{dW}$

$$\frac{d\sigma}{dW} = 2\pi \frac{e^4}{mc^2} \frac{1}{\beta^2 W^2} \left( 1 - \beta^2 \frac{W}{W_{\max}} \right)$$

semi-classical (ie. when  $W \ll W_{\max}$ )

→ technically, this is expression for spin 0 particles

→ when  $W \ll$  incident energy, spin dependence goes away

For massive particle

(33)

$$\frac{dE}{dx} = \frac{4\pi n_e}{m\nu^2} z^2 e^4 \left( \ln \left( \frac{2m\nu^2 \gamma^2}{I} \right) - \beta^2 \right)$$

### Bethe-Bloch

- depends on

- velocity of incident particle
- material electron density,  $n_e$
- $I$  = mean ionization potential of material (or, excitation)

$I$  is main parameter

→ the average orbital frequency,  $\nu$   
times Planck constant

$$\underline{I = h\nu \sim 16 Z^{0.9} \text{ eV} \quad (Z > 1)}$$

→ the value of  $\nu$  is a weighted average of frequencies for atomic levels

- weighted by "oscillator strengths" (mainly from empirical measurement)

→ some variation as move thru periodic table

# Behavior of Energy Loss

(34)

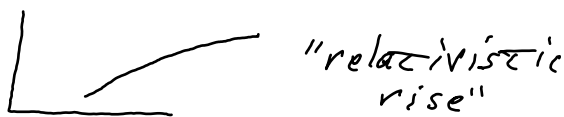
$$\frac{dE}{dx} \propto \frac{1}{\beta^2} \ln A\beta^2\gamma^2$$

↳ const.

For low velocities ( $\beta\gamma < 3$ )

$$\frac{dE}{dx} \propto 1/\beta^2$$


When  $\beta\gamma > 4$ ,  $dE/dx \propto \ln\beta^2\gamma^2$



Minimum ionizing

→ condition when  $\beta\gamma \sim 3-4$

→ all incident particles roughly

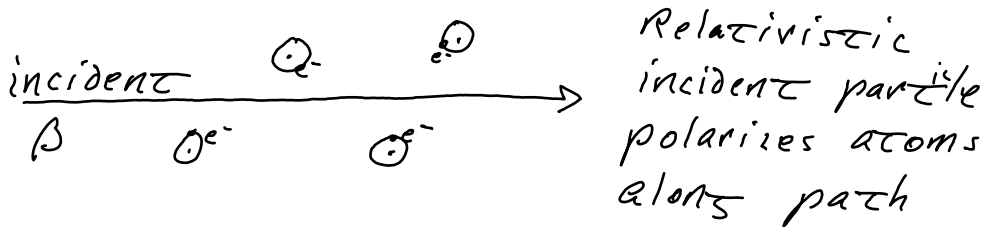
same  $dE/dx \sim 2 \text{ MeV/g cm}^2$

(note: the overall structure here has similarity to semi-classical picture.)

# Density and Fermi Plateau

(35)

We've considered target atoms as if they are each isolated  
- for dense materials



$\therefore$  atomic electrons furthest from path see reduced  $\vec{E}$  field  
(~~prop beam in~~ screened)

- they contribute less ionization than expected

As incident particle  $\gamma$  rises:

$\lambda_{\text{max}} \Rightarrow \gamma v / v \rightarrow$  larger volume of integration

(in quantum case, this relates to small momentum transfer)

$\therefore$  density effect larger

$\frac{dE}{dx} \propto \ln \gamma^2 \rightarrow \ln \gamma \rightarrow$  constant

Fermi plateau

(36)

Bethe-Bloch modified to

$$\frac{dE}{dx} = \frac{4\pi n_e z^2 e^4}{m v^2} \left[ \ln \frac{2m v^2 \gamma^2}{I} - \beta^2 - \frac{\delta(\gamma)}{2} \right]$$

 $\delta(\gamma)$  = density parameter

- in terms of atomic properties

# Fluctuations in Energy Loss (37)

$$\frac{d\sigma}{dW} \propto \frac{1}{W^2} \quad \therefore \text{small energy transfers most likely}$$

- the  $\frac{dE}{dx}$  calculated (Bethe-Block)
- ~~most~~ average value
- most probable energy transfers lower  $E$
- high tail from  $\delta$ -rays

Number  $\delta$ -rays produced with energy  $> E_1$  in thickness  $x$

$$N(E \geq E_1) = \frac{2\pi n_e z^2 e^4}{m\nu} x \left( \frac{1}{E_1} - \frac{1}{E_{\max}} \right)$$

maximum possible energy transfer

# Electrons & Positrons

(38)

- Ionization not primary energy loss mechanism for  $e^\pm$  (in most cases)

→ somewhat different than for other particles

$$\frac{d\sigma}{dw}(\mathcal{E}, w) = \frac{2\pi r_e^2}{mv^2} \left[ \frac{1}{w^2} - \frac{1}{w(\mathcal{E}-w)} \times \right.$$

$$\left. \frac{mc^2(2\mathcal{E}+mc^2)}{(\mathcal{E}+mc^2)^2} \right]$$

$\mathcal{E} = K.E.$  of incident  $e^\pm$

$$+ \frac{1}{(\mathcal{E}-w)^2} + \frac{1}{(\mathcal{E}+mc^2)^2} \Bigg]$$

Moller  
cross-section

(Bhabha for  $e^+$ )

$$\frac{d\mathcal{E}}{dx} = \frac{2\pi r_e^2 e^4}{mc^2} \left( 2 \ln \frac{2mc^2}{I} + (3) \ln \gamma - (1.95) \right)$$

= 2 for  $e^+$

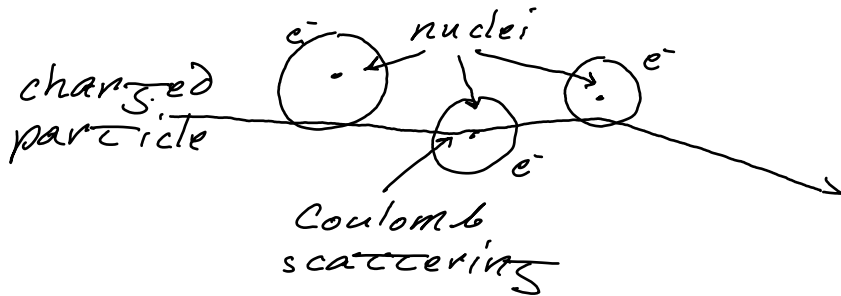
for electrons

(= 4 for  $e^+$ )

- relativistic rise smaller than for other particles (i.e.  $\propto \ln \gamma$ )

# Elastic Scattering

(39)



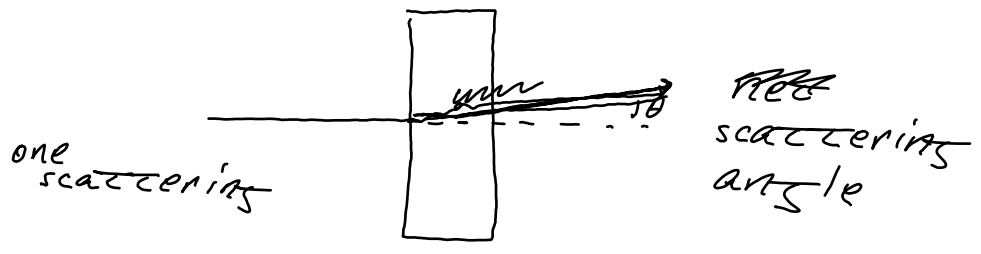
- Deflections of particle trajectory
- nuclei +  $e^-$  ~~involved~~ involved
  - nuclei dominate this process

Make some assumptions:

- no energy loss
- only a change charged particle trajectory results



Consider a particle  
 → charge  $z$ , mass  $m$   
 - incident on a block of material



Charged particles in medium receive  $\perp$  change in momentum  
 → incident particle gets opposite

(one scattering)  $p_{\perp} = \frac{2z^2 e^2}{bv}$  (from earlier)

This means

$$\theta = \frac{p_{\perp}}{p} = \frac{2z^2 e^2}{bv p}$$

Or

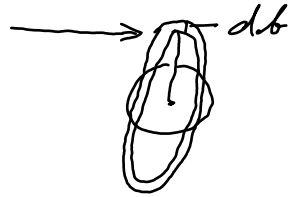
$$\underline{b = \frac{2z^2 e^2}{bv \theta}}$$

# Differential Cross Section

(41)

Want to know cross section for scattering between  $\theta$  and  $\theta + d\theta$

$$d\sigma = 2\pi b db$$



Substituting for  $b$  and  $db$ , we have

$$d\sigma = 2\pi \left( \frac{2Zze^2}{pv} \right)^2 \frac{d\theta}{\theta^3}$$

For small angles,  $d\Omega \approx 2\pi \theta d\theta$

$$\therefore \boxed{\frac{d\sigma}{d\Omega} = 2^2 Z^2 r_e^2 \left( \frac{mc}{\beta p} \right)^2 \frac{1}{(\theta/2)^4}}$$

spin  $\neq$

Full Rutherford form:

$$\theta/2 \rightarrow \sin(\theta/2)$$

(spin  $1/2$ : multiply by  $(1 - \beta^2 \sin^2 \theta/2)$ )

# Total Elastic Cross Section

(42)

Need to integrate

→ Rutherford breaks down when too far from atom

- nuclear charge screened by atomic electrons

$$\therefore \theta^2/4 \rightarrow (\theta^2 + \theta_{\min}^2)/4$$

↳ cross section levels at very small angles

Estimating  $\theta_{\min}$ :

By atomic radius  $\sim b$

$$\theta'_{\min} \approx 2 \cdot 2^{4/3} \frac{mc^2}{p\hbar} \left(\frac{r_e}{a_0}\right)$$

→ use the

By uncertainty principle

$$\theta_{\min} = \alpha \cdot 2^{1/3} \frac{mc}{p}$$

→ smaller one

Also a  $\theta_{\max}$  when  $\lambda \sim$  nuclear radius

$$\theta_{\max} \approx \frac{2A^{-1/3} mc}{\alpha p}$$

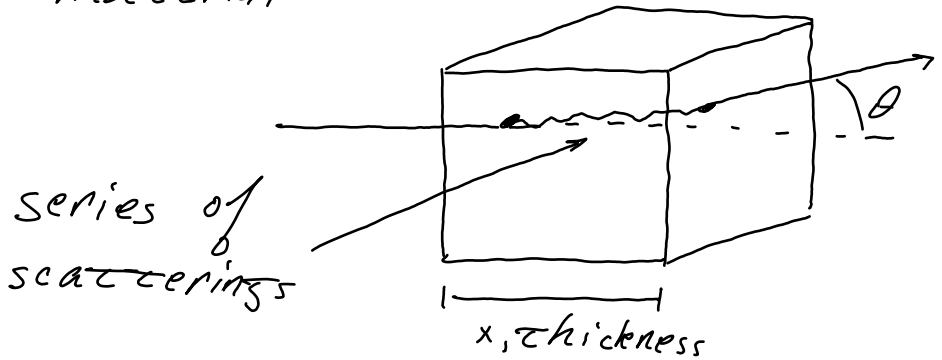
These give us

$$\sigma_{el.} = \int_{\theta_{\min}}^{\theta_{\max}} \frac{d\sigma}{d\Omega} \sin\theta d\theta d\phi = \boxed{\pi r_e^2 4 \left(\frac{2 \cdot 2^{2/3}}{\alpha \beta}\right)^2}$$

# Multiple Scattering

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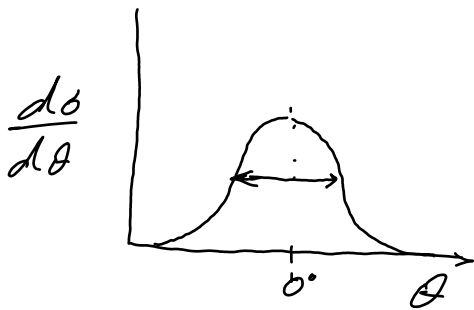
particle passing thru block of material



→ average deflection of each scattering  $0^\circ$

→ but a dispersion in outgoing particle net angle

$n$   
 $w_{sc}$  is  
typical  
value of  
 $|\theta|$ ?



Calculate average  $\theta^2$

(49)

$$\theta_s^2 = x n_a \int_{\theta_{\min}}^{\theta_{\max}} \theta^2 \frac{d\sigma}{d\Omega} d\Omega$$

use Rutherford

statistical approach

Abce:

$\theta_s = \text{root-mean squared } \theta$

$$\theta_s \propto \sqrt{n_a}$$

$$= \sqrt{\langle \theta^2 \rangle} = 8\pi x n_a r_e^2 z^2 \left(\frac{mc}{\beta p}\right)^2 \ln \frac{2}{2^2 A^{1/3} z^{2/3}}$$

this can be expressed (if  $A=22$ )

$$\theta_s = \frac{\sqrt{4\pi} z mc^2}{\beta p} z \sqrt{x/\chi_0}$$

The 'radiation length',  $\chi_0$  is distance over which a significant energy is lost due to bremsstrahlung.

$$\chi_0 = \frac{A}{4\alpha n_a z^2 r_e^2 \ln(183/z^{2/3})}$$

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Example: an electron ( $z=1$ )  
of 16 MeV energy passes thru  
 $20 \text{ mg/cm}^2$  of iron ( $\chi_0 = 13.8 \text{ g/cm}^2$ )

$$\underline{\underline{\theta_s}} \sim \frac{13.6 \text{ MeV}}{v\rho} \sqrt{\frac{x}{\chi_0}} \quad (v = \beta c \sim 1)$$

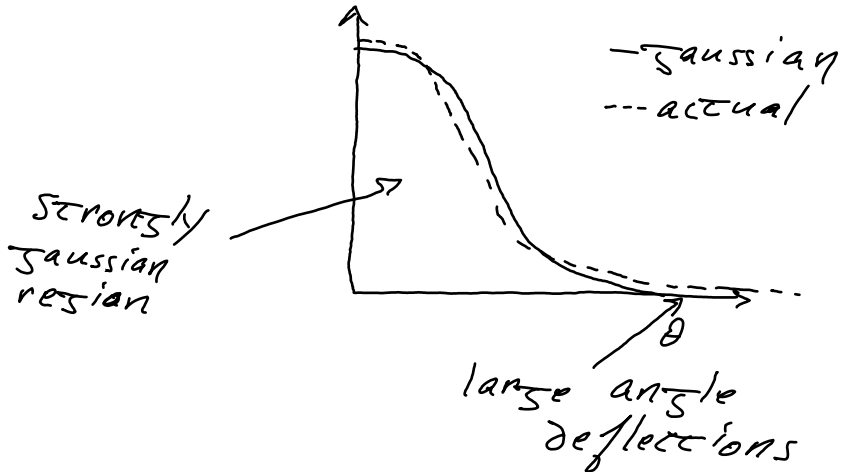
$$= \frac{13.6 \text{ MeV}}{16 \text{ MeV}} \sqrt{\frac{20 \text{ mg/cm}^2}{13.8 \text{ g/cm}^2}}$$

$$= \underline{\underline{0.031 \text{ rad}}}$$

# Gaussian Behavior

(4/6)

Scattering angle obeys a rough gaussian shape



large angle tail

- most non-gaussian
- generally from one large angle deflection in material
- these obey Rutherford formula