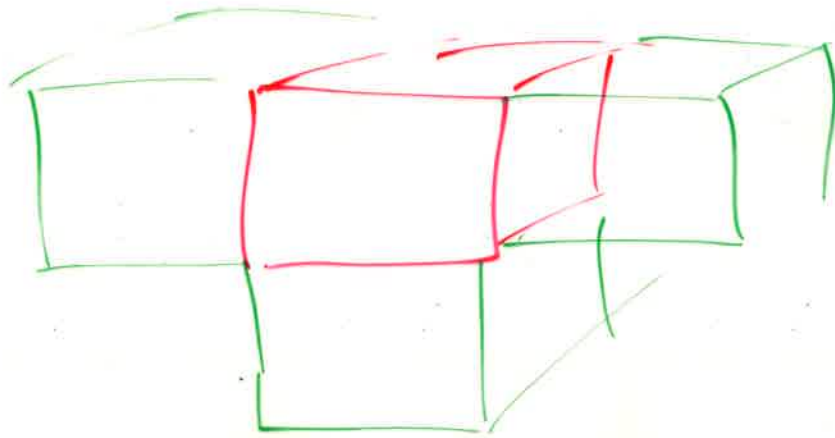


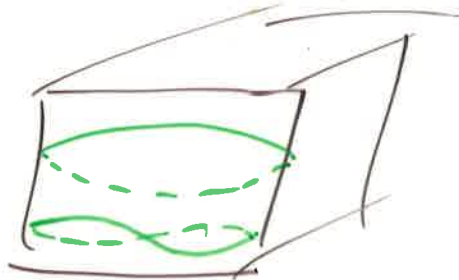
# Periodic Boundary Conditions



$$\int \frac{d^3k}{(2\pi)^3}$$

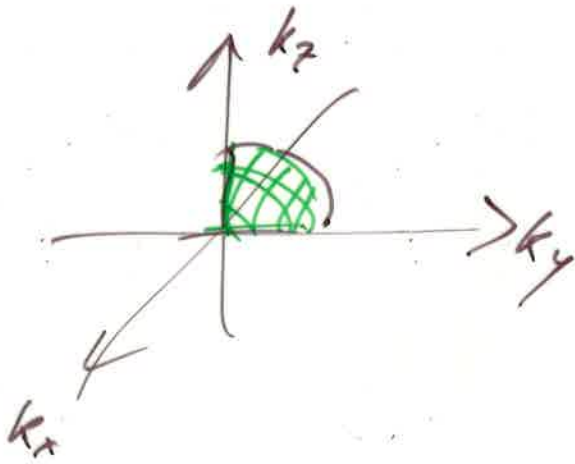
integrate  
over  
all  $k$

Hard wall B.C.



$$\frac{1}{8} \int \frac{d^3k}{(\pi)^3}$$

integrate over  
first octant



Heat capacity

$$C_v = \left( \frac{\partial U}{\partial T} \right)_v = \frac{32 \pi^5 k_B^4 T^3 V}{15 h^3 c^3}$$

$$\lim_{T \rightarrow 0} C_v \Rightarrow 0 \quad (\text{Third Law})$$

Entropy

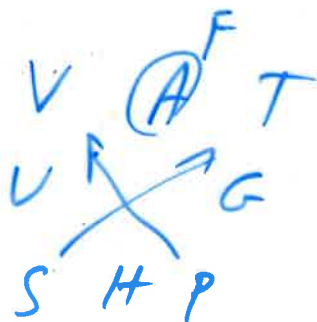
$$S = \int_0^T \frac{C_v(T)}{T} dT = \frac{32 \pi^5 k_B^4 T^3 V}{45 h^3 c^3}$$

Pressure

1st Law  $dU = T dS - P dV + \mu dN$

$$dS = \frac{dU}{T} - \frac{P}{T} dV + \frac{\mu}{T} dN$$

$$P = T \left( \frac{\partial S}{\partial V} \right)_{UN} = \frac{\frac{32 \pi^5 k_B^4 T^4}{45 h^3 c^3}}{\times} \text{ wrong}$$



Pressure

$$dF = -SdT - pdV + \mu dN$$

$$F = U - TS \leftarrow$$

$$P = - \left( \frac{\partial F}{\partial V} \right)_{TN}$$

$$F = -k_B T \ln(Z)$$

$$F = U - TS = aVT^4 - \frac{4}{3}aVT^4 = -\frac{1}{3}aVT^4$$

$$-\left( \frac{\partial F}{\partial V} \right)_{TN} = +\frac{1}{3}aT^4 = \frac{1}{3} \frac{U}{V} = \frac{1}{3} a \quad \checkmark$$

Number of Photons

1.20...

Riemann zeta function

$$N = 16\pi \zeta(3) \frac{V T^{3/3}}{h^3 c^3}$$

$$\int_{x=0}^{\infty} \frac{x^2}{e^x - 1} dx = 2 \zeta(3)$$

Planck distribution solves the UV catastrophe.

$$\text{Energy Density} = u = \frac{U}{V} = \frac{1}{\pi^2} \int_{k_r=0}^{\infty} dk_r \frac{k_r^2 \hbar c k_r}{e^{\beta \hbar c k_r} - 1}$$

$\nu$  = linear frequency

$$\hbar c k = \hbar \omega = h \nu$$

$$[\omega] = \frac{\text{rad}}{\text{sec}}$$

$$[\nu] = \frac{1}{\text{sec}} = \text{Hz}$$

$$\omega = 2\pi \nu$$

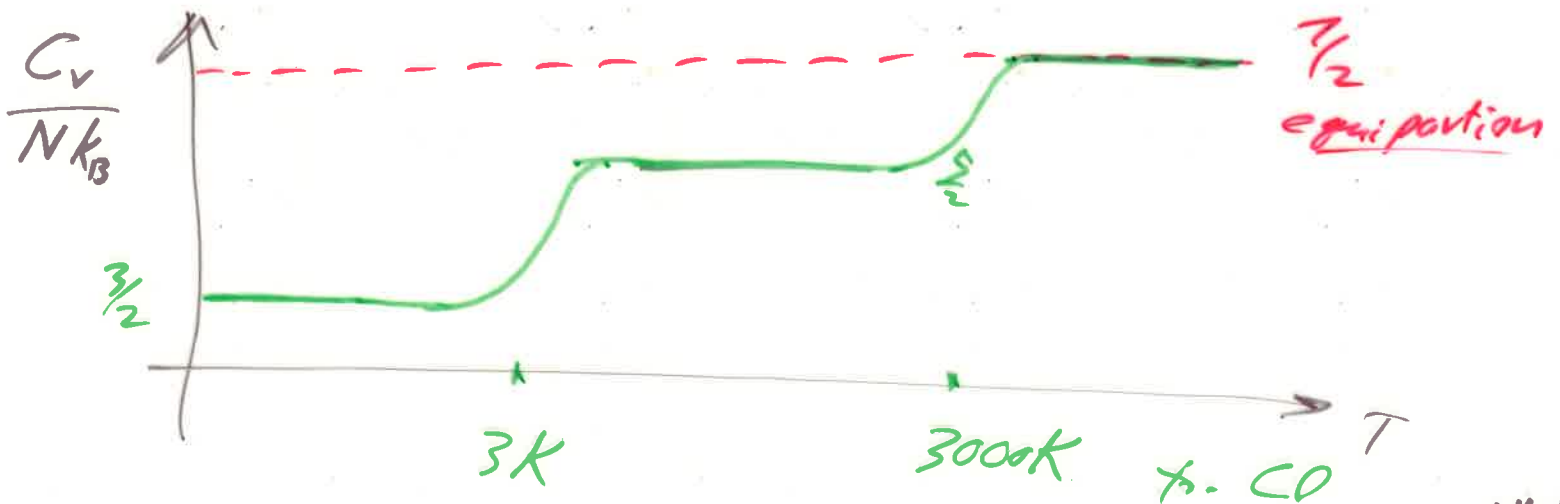
$$u(\nu) = \frac{1}{\pi^2 c^2} \frac{h \nu^3}{e^{\beta h \nu} - 1}$$

$u(\nu)$

small  $\nu \ll 1$   $e^{\beta h \nu} = 1 + \beta h \nu + \dots$   
 $u(\nu) \sim \nu^2$

large  $\nu \gg 1$   $u(\nu) \sim e^{-\nu}$

② Heat capacity for diatomic gas



# Classical Canonical Partition function

$$Z(T, V, N) = \frac{1}{N!} z_1^N \leftarrow \text{one molecule}$$

$\sim$  to avoid Gibbs paradox

$$z_1 = \int \frac{d^3 p_1 d^3 p_2 d^3 r_1 d^3 r_2}{h^6} \exp \left[ -\beta \left\{ \frac{\vec{p}_1^2}{2m_1} + \frac{\vec{p}_2^2}{2m_2} + V(\vec{r}_1 - \vec{r}_2) \right\} \right]$$

$$\vec{p}_i^2 \equiv \vec{p}_i \cdot \vec{p}_i = p_{ix}^2 + p_{iy}^2 + p_{iz}^2$$

↑  
potential energy

Define  $\vec{R}_{cm} = \frac{m_1 \vec{r}_1 + m_2 \vec{r}_2}{m_1 + m_2}$

$$\vec{P}_{cm} = \vec{p}_1 + \vec{p}_2$$

Center of Mass  
Coord's

$$M = m_1 + m_2$$

Relative coordinate

$$\vec{r} = \vec{r}_1 - \vec{r}_2$$

$$\vec{p} = \frac{m_2 \vec{p}_1 - m_1 \vec{p}_2}{m_1 + m_2}$$

$$\mu = \frac{m_1 m_2}{m_1 + m_2} \quad \text{reduced mass}$$

$$z_1 = \underbrace{\int \frac{d^3 P_{cm} d^3 R_{cm}}{h^3} e^{-\beta \frac{\vec{P}_{cm}^2}{2M}}}_{\text{translation}} \cdot \underbrace{\int \frac{d^3 p d^3 r}{h^3} e^{-\beta \frac{\vec{p}^2}{2\mu} - \beta V(\vec{r})}}_{\text{vibrations + rotations}}$$