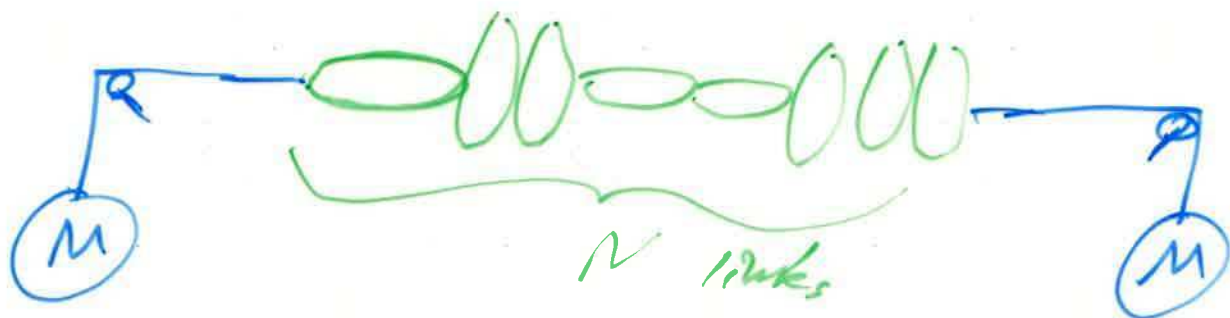
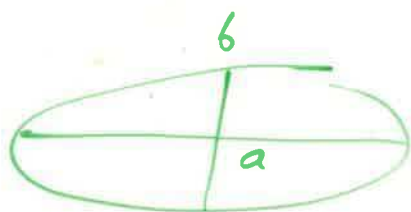


One-dimensional chain w/ masses & link



Tension in chain? $\tau = Mg$



At temp T , what is L ?

Energy states of one link.



$$E_1 = -\tau a$$



$$E_2 = -\tau b$$

Partition function for one link

$$z = e^{-\beta(-\tau a)} + e^{-\beta(-\tau b)} = e^{\beta \tau a} + e^{\beta \tau b}$$

$$Z = z^N = (e^{\beta \tau a} + e^{\beta \tau b})^N$$

$$\ln(Z) = N \ln(e^{\beta \tau a} + e^{\beta \tau b})$$

What therm. potential should we calculate

$$U = -\frac{\partial}{\partial \beta} \ln(Z) ; F = -\frac{1}{\beta} \ln(Z)$$

$$\text{Expect } \langle L \rangle = \mp \frac{\partial (\text{Potential})}{\partial (\text{variable})} \Big|_{x,y}$$

$$dF = -SdT + \tau dL + \mu dN^0$$

\Rightarrow

$$dF = \left(\frac{\partial F}{\partial T} \right)_{LN} dT + \left(\frac{\partial F}{\partial L} \right)_{TN} dL + (-)$$

$$\Rightarrow S = -\left(\frac{\partial F}{\partial T} \right)_{LN} , \tau = \left(\frac{\partial F}{\partial L} \right)_{TN} \quad \boxed{F + PV = G}^{\text{gas}}$$

but we want $\langle L \rangle$, not $\langle \tau \rangle$ $F - L\tau = G$
(Legendre Transformation)

$$dG = -SdT - Ld\tau$$

$$U - TS = F$$
$$U - TS + PV = G$$

$$dG = \left(\frac{\partial G}{\partial T} \right)_{\tau, N} dT + \left(\frac{\partial G}{\partial \tau} \right)_{T, N} d\tau + (-)$$

$$\Rightarrow \langle L \rangle = -\left(\frac{\partial G}{\partial \tau} \right)_{T, N}$$

How do we get G ? - Already have it.

Z is not the canonical partition function.

$-\tau_a, -\tau_b$ are not E_1 & E_2

$-\tau_a = H_1, -\tau_b = H_2$ enthalpies

script Z

\downarrow
 $\mathcal{Z} = e^{-\beta H_1} + e^{-\beta H_2} \leftarrow$ Gibbs partition function

$\rightarrow G = -\frac{1}{\beta} \ln(\mathcal{Z}) = -k_B T \ln(\mathcal{Z})$
 \uparrow Gibbs

$\rightarrow F = -k_B T \ln(Z)$
 \uparrow canonical

$Q = \sum_i e^{-\beta(E_i - \mu N_i)} \leftarrow$ grand partition function
 \uparrow Gibbs sum

$\rightarrow \Phi = -k_B T \ln(Q)$

\uparrow grand potential

\mathcal{Z} - script G

Ω -

$$\frac{dZ}{Z} = (e^{+\beta \epsilon_a} + e^{+\beta \epsilon_b})^N$$

$$G = -k_B T \ln(Z) = -k_B T N \ln(e^{\beta \epsilon_a} + e^{\beta \epsilon_b})$$

$$\langle U \rangle = - \left(\frac{\partial G}{\partial \tau} \right)_{T, N} = - \frac{\partial}{\partial \tau} \left[-k_B T N \ln(e^{\beta \epsilon_a} + e^{\beta \epsilon_b}) \right]$$

$$\langle U \rangle = \frac{\cancel{k_B T N} [a e^{\beta \epsilon_a} + b e^{\beta \epsilon_b}]}{e^{\beta \epsilon_a} + e^{\beta \epsilon_b}}$$

$$\langle U \rangle = \frac{N [a e^{\beta \epsilon_a} + b e^{\beta \epsilon_b}]}{e^{\beta \epsilon_a} + e^{\beta \epsilon_b}}$$

Low T , High β : Na ✓

High T , Low β : $\frac{N(a+b)}{2}$ ✓

Find the chemical potential μ for an ideal gas.

$$Z = \left[\frac{eV}{N} \left(\frac{\sqrt{2\pi m k_B T}}{h} \right)^3 \right]^N \quad \text{cf. Lect 16}$$

See Schroder p. 118 for the hard way.

Get a potential: $F = -k_B T \ln(Z)$

$$dF = -SdT - pdV + \underline{\mu} dN$$

$$= \left(\frac{\partial F}{\partial T} \right)_{VN} dT + \left(\frac{\partial F}{\partial V} \right)_{TN} dV + \left(\frac{\partial F}{\partial N} \right)_{TV} dN$$

$$\mu = \left(\frac{\partial F}{\partial N} \right)_{TV} = -k_B T \frac{\partial}{\partial N} \left\{ N \ln \left[\frac{eV}{N} \left(\frac{\sqrt{2\pi m k_B T}}{h} \right)^3 \right] \right\}_{TV}$$

Define:

Thermal de Broglie Wavelength $\lambda_Q \equiv \frac{h}{\sqrt{2\pi m k_B T}}$

in analogy with the de Broglie wavelength

$$\lambda = \frac{h}{p} = \frac{h}{\sqrt{2mE}}$$

further, define the quantum volume $V_Q \equiv \lambda_Q^3$

intensive volume per "particle"

$$\mu = -k_B T \frac{\partial}{\partial N} \left\{ N \ln \left[\frac{eV}{N v_Q} \right] \right\}_{TV}$$

$$= -k_B T \left\{ \ln \left[\frac{eV}{N v_Q} \right] + N \frac{\partial}{\partial N} [\cancel{\ln e} + \cancel{\ln V} - \ln N - \cancel{\ln v_Q}] \right\}_{TV}$$

$$= -k_B T \left\{ \ln \left[\frac{eV}{N v_Q} \right] + N \left[-\frac{1}{N} \right] \right\}$$

$$= -k_B T \left\{ \underline{\ln e} + \ln \left[\frac{V}{N v_Q} \right] - \underline{1} \right\}$$

$$= -k_B T \ln \left[\frac{V}{N v_Q} \right]$$

Schröder Eq. 3.63

Call $\frac{V}{N} = v$ ($\frac{\text{extensive}}{\text{extensive}} = \text{intensive}$)

$$\mu = k_B T \ln \left(\frac{v_Q}{v} \right) \quad \text{for ideal gas}$$

