

one molecule partition function

$$Z_{qm}^{rot} = \sum_{l=0}^{\infty} (2l+1) e^{-\frac{\beta \hbar^2}{2I} l(l+1)}$$


high T , (low β): sum \rightarrow integral

$$Z_{qm}^{rot} \rightarrow \int_{l=0}^{\infty} dl (2l+1) e^{-\frac{\beta \hbar^2}{2I} l(l+1)}$$

$$x = l(l+1) \quad dx = (2l+1) dl$$

$$Z_{qm}^{rot} = \int_{x=0}^{\infty} dx e^{-\frac{\beta \hbar^2}{2I} x} = \frac{2I}{\beta \hbar^2} e^{-\frac{\beta \hbar^2}{2I} x} \Big|_{x=0}^{\infty}$$

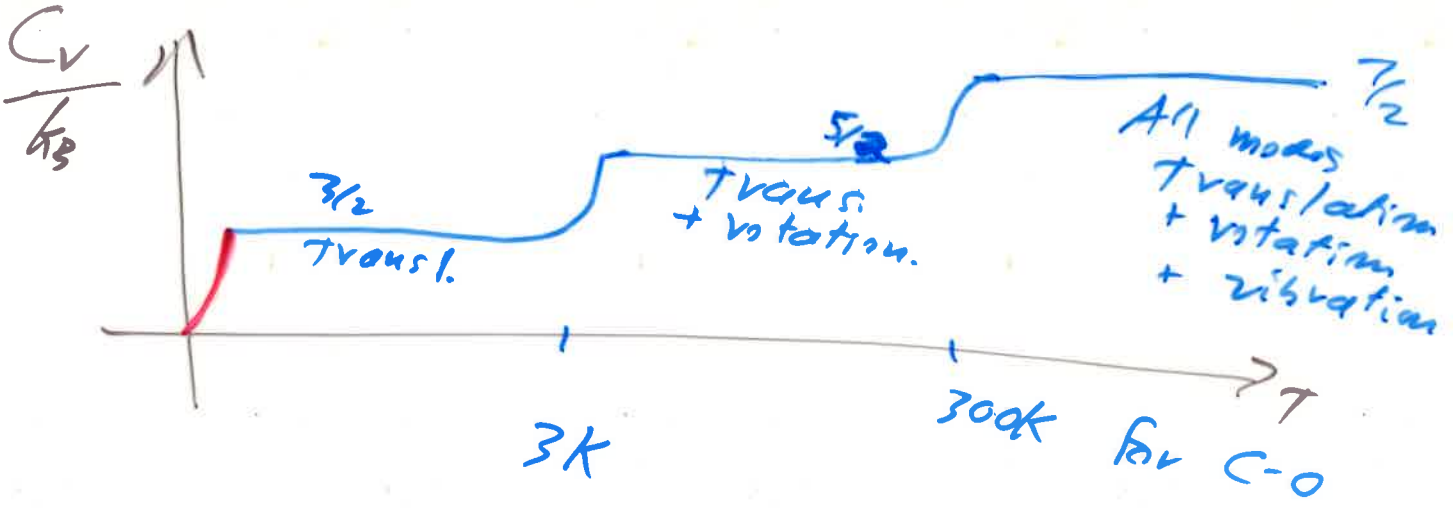
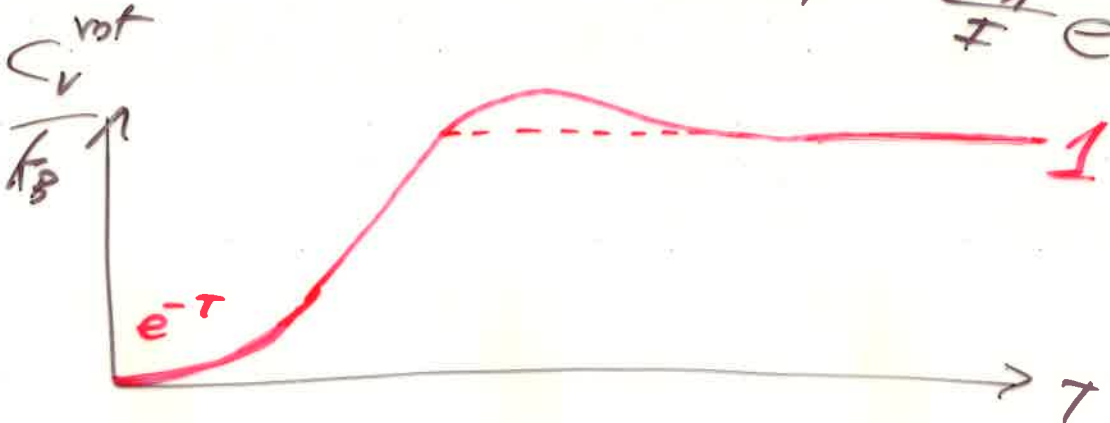
$$= \frac{2I}{\beta \hbar^2} - \text{reproduces classical } Z^{rot} \checkmark$$

Low T , high β - non-identical atoms in molecule
eg. C-O 

$$Z_{qm}^{rot} = \underbrace{1}_{l=0} + \underbrace{3 e^{-\frac{\beta \hbar^2}{2I} 2}}_{l=1} + \dots$$

$$\epsilon_{rot} = -\frac{\partial \ln(Z)^{rot}}{\partial \beta}$$

$\beta \ll 1 \rightarrow k_B T$
 $\beta \gg 1 \rightarrow \frac{3/2}{T} e^{-\frac{\beta \hbar^2}{I}} + \dots$



Complications for identical atoms

O_2 spin-0 bosons rotation by π gives the same state back.

H_2 spin-1/2 fermions high T, low $\beta \rightarrow \frac{3}{2}$

D_2 spin-1 bosons $-\frac{\partial}{\partial \beta} (\ln Z^{rot} - \ln(2)) = \epsilon_{rot}$

$C_v = \frac{d\epsilon}{dT} = \text{same as w/o } 2!$

Oxygen
 O_2

- overall wavefunction is symmetric (even)

$$\Psi = \Psi_{spin} \cdot \Psi_{orbital}$$

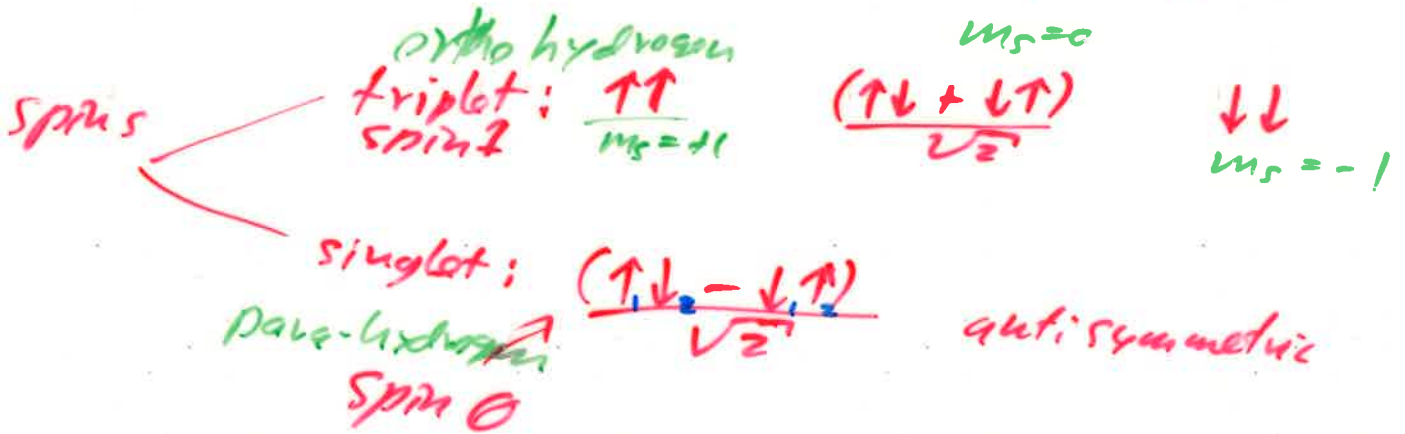
$$+1 = \underbrace{\text{even}}_{+1} \cdot \underbrace{(-1)^l}_{+1}$$

$$(-1)^l = +1 \rightarrow \text{even } l \text{ only.}$$

hydrogen

H_2 fermions

overall wavefunction is odd = antisymmetric



$$\Psi_{overall} = \Psi_{spin} \cdot \Psi_{orbital}$$

$$(-1) = \begin{cases} (-1) \cdot (+1) \rightarrow \text{even } l \text{ para} \\ (+1) \cdot (-1) \rightarrow \text{odd } l \text{ ortho} \end{cases}$$

$$(-1) \cdot (-1) \rightarrow \text{even } l \text{ ortho}$$

D_2 deuterium
- spin 1 - spin bosons

Ψ_{overall} is symmetric (even)

$$\Psi_{\text{overall}} = \Psi_{\text{spin}} \cdot \Psi_{\text{orbital}}$$

$$(+1) = ? \quad ?$$

