

This is  
OK

$$P = - \left( \frac{\partial U}{\partial V} \right)_{S, N}$$

$$= - \left[ \underbrace{\left( \frac{\partial U}{\partial V} \right)_{T, N}}_0 - \underbrace{\left( \frac{\partial U}{\partial S} \right)_{V, N}}_T \cdot \left( \frac{\partial S}{\partial V} \right)_{T, N} \right]$$

S = ?  
↓

Entropy of the ideal gas

$$S = k_B \ln(\# \text{ microstates}) = \infty \dots$$

Phase space is chunky  $\delta x \delta p \sim h$

$$Z = \frac{V^N}{N!} \left( \frac{2\pi m}{\beta} \right)^{\frac{3N}{2}} \left( \frac{1}{h^3} \right)^N \rightarrow \left[ \frac{eV}{N} \left( \frac{2\pi m}{h^2 \beta} \right)^{\frac{3}{2}} \right]^N$$

$$S = - \left( \frac{\partial F}{\partial T} \right)_{V, N} = + \frac{\partial}{\partial T} \left[ k_B T \ln(Z) \right]_{V, N}$$

$$S = k_B \ln(Z) + N k_B T \frac{\partial}{\partial T} \left[ \ln \left( \frac{eV}{N} \left( \frac{2\pi m}{h^2 \beta} \right)^{\frac{3}{2}} \right) \right]_{V, N}$$

$$\frac{\partial}{\partial T} \left[ \ln(e) + \ln(V) - \ln(N) + \frac{3}{2} \ln\left(\frac{2\pi m k_B}{h^2}\right) + \frac{3}{2} \ln(T) \right]_{VN}$$

$$= \frac{3}{2} \frac{1}{T}$$

Classical  
limit  
 $h \rightarrow 0$

$$S = N k_B \ln \left[ \frac{eV}{N} \left( \frac{2\pi m}{h^2 \beta} \right)^{3/2} \right] + \frac{3}{2} N k_B$$

$$= N k_B \left\{ \ln(e) + \ln \left[ \frac{V}{N} \left( \frac{2\pi m}{h^2 \beta} \right)^{3/2} \right] \right\} + \frac{3}{2} N k_B$$

$$= N k_B \left\{ \ln \left[ \frac{V}{N} \left( \frac{2\pi m}{h^2 \beta} \right)^{3/2} \right] + \frac{5}{2} \right\}$$

Sackur-Tetrode  
monatomic ideal gas

$\uparrow$  from  $\frac{1}{N!}$   $\times$  Avoids the Gibbs paradox

$$\left( \frac{\partial S}{\partial V} \right)_{TN} = \frac{N k_B}{V}$$

$$S = N k_B \left\{ \ln(V) - \ln(N) + \frac{3}{2} \ln\left(\frac{2\pi m}{h^2 \beta}\right) + \frac{5}{2} \right\}$$

$$\left( \frac{\partial S}{\partial V} \right)_{TN} = \frac{N k_B}{V}$$

$$P = - \left[ 0 - T \left( \frac{N k_B}{V} \right) \right] \Rightarrow PV = N k_B T \checkmark$$

$$S = N k_B \left\{ \ln(V) - \ln(N) + \frac{3}{2} \ln\left(\frac{2\pi m}{h^2 \beta}\right) + \frac{5}{2} \right\}$$

# Chemical Potential for an Ideal Gas

$$Z = \left[ \frac{eV}{N} \left( \frac{\sqrt{2\pi m k_B T}}{h} \right)^3 \right]^N$$

Get a potential :  $F = -k_B T \ln(Z)$

$$dF = -S dT + \mu dN - p dV$$

$$\mu = \left( \frac{\partial F}{\partial N} \right)_{TV} \quad dF = \left( \frac{\partial F}{\partial T} \right)_{N,V} dT + \left( \frac{\partial F}{\partial N} \right)_{T,V} dN + \left( \frac{\partial F}{\partial V} \right)_{T,N} dV$$

$$\mu = \frac{\partial}{\partial N} \left\{ -k_B T N \left[ \ln(e) + \ln(V) - \ln(N) + \ln \dots \right] \right\}$$

$$= -k_B T \ln \left( \frac{Ve}{N} \left( \frac{\sqrt{2\pi m k_B T}}{h} \right)^3 \right)$$

$$+ k_B T N \left[ \frac{\partial F}{\partial N} \ln(e) + \ln(V) - \ln(N) + \ln \dots \right]$$

$$\mu = -k_B T \left[ \ln \left( \frac{Ve}{N} \left( \frac{\sqrt{2\pi m k_B T}}{h} \right)^3 \right) - 1 \right]$$

extensive  $\rightarrow V = v \leftarrow$  intensive  $\frac{1}{3} \frac{1}{\omega}$   
 extensive  $\rightarrow N$

Define:  $\lambda_Q \equiv \frac{h}{\sqrt{2\pi m k_B T}}$  thermal de Broglie wavelength

In analogy with de Broglie wavelength

$$\lambda = \frac{h}{p} = \frac{h}{\sqrt{2mE}}$$

Define: Quantum volume  $v_Q \equiv \lambda_Q^3$   
 intensive

$$\mu = -k_B T \ln\left(\frac{v}{v_Q}\right)$$

