

# Quantum Mechanics for Rotations (1920s)

$$E_{QM}^{\text{rot}} = \frac{\hbar^2}{2I} l(l+1)$$

$l = 0, 1, 2, 3, \dots$   
quantum number

$g_l = (2l+1)$  degeneracy

$$Z_{QM}^{\text{rot}} = \sum_{l=0}^{\infty} (2l+1) e^{-\beta \frac{\hbar^2}{2I} l(l+1)}$$

high  $T$ , low  $\beta$ : sum  $\rightarrow$  integral

$$Z_{QM}^{\text{rot}} \underset{\text{high } T}{=} \int_{l=0}^{\infty} dl (2l+1) e^{-\beta \frac{\hbar^2}{2I} l(l+1)}$$

$$x = l(l+1)$$

$$dx = (2l+1) dl$$

$$= \int_{x=0}^{\infty} dx e^{-\frac{\beta \hbar^2}{2I} x} = \frac{-2I}{\beta \hbar^2} e^{-\frac{\beta \hbar^2}{2I} x} \Big|_{x=0}^{\infty}$$

$$= \frac{2I}{\beta \hbar^2}$$

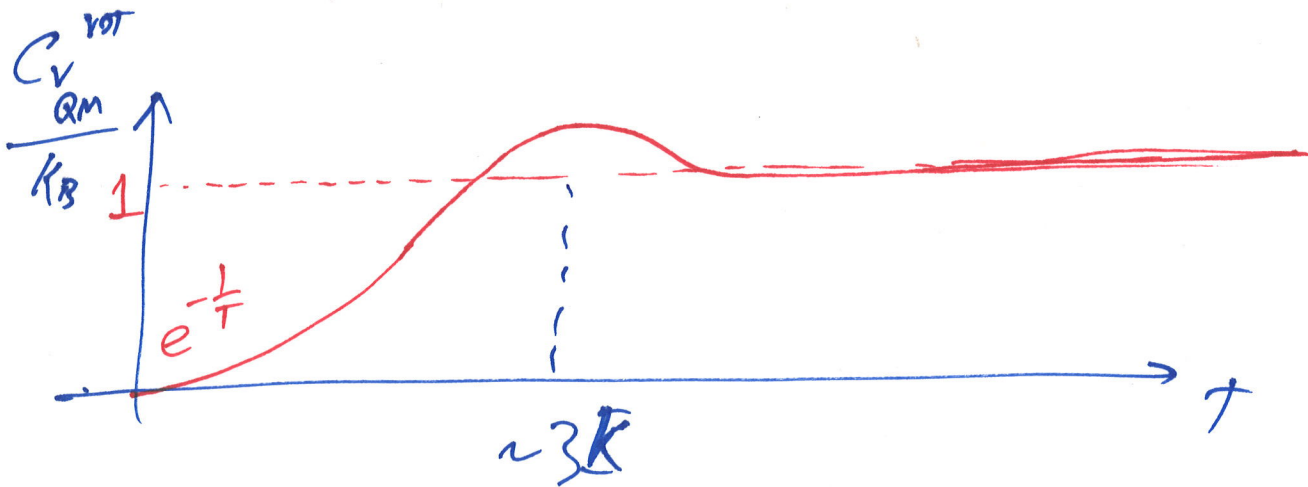
reproduces the classical result.

Low  $T$ , high  $\beta$  Non-identical nuclei, e.g. CO


$$z_{QM}^{\text{rot}} = \underbrace{1}_{l=0} + \underbrace{3e^{-\frac{\beta \hbar^2}{2I}}}_l + \underbrace{5e^{-\frac{\beta \hbar^2}{2I} 6}}_{l=2} + \dots$$

$$E_{QM}^{\text{rot}} = \frac{-\partial \ln(z_{QM}^{\text{rot}})}{\partial \beta}$$

→ high  $T$  low  $\beta$   $k_B T$   
 → low  $T$  high  $\beta$   $\frac{3\hbar^2}{I} e^{-\frac{\beta \hbar^2}{I}} + \dots$



Complications for identical atoms

$O_2$   spin-0 bosons

$H_2$   spin-1/2 fermions

$D_2$   spin-1 bosons

rotation by  $\pi$  radians gives the same state back

got even or odd  $l$  terms in the sum.

oxygen  $O_2$  spin-0 bosons  $\Rightarrow$  total wavefunction is symmetric

$$\Psi = \Psi_{spin} \cdot \Psi_{orbital}$$

↑ sym

total wavefunction

$l$  is quantum number

$$\hat{L}^2 \Psi_{orb} = l(l+1)\hbar^2 \Psi_{orb}$$

$$l \in \{0, 1, 2, \dots\}$$

$\Rightarrow \Psi_{orb}$  must be symmetric  $\Rightarrow$   
 $\Rightarrow l$  is even

hydrogen  $H_2$  spin- $\frac{1}{2}$  fermions  $\Rightarrow$  total wavefunction is antisymmetric

$$\Psi_{total} = \Psi_{spin} \cdot \Psi_{orbital}$$

↑ antisym

spins  $\left\{ \begin{array}{l} \text{triplet spin-1} \\ \text{singlet spin-0} \end{array} \right.$

ortho hydrogen

$(\uparrow\uparrow)_{12}$  or  $\frac{(\uparrow_1\downarrow_2 + \downarrow_1\uparrow_2)}{\sqrt{2}}$  or  $(\downarrow\downarrow)_{12}$

spin symmetric

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para hydrogen

$\frac{(\uparrow_1\downarrow_2 - \downarrow_1\uparrow_2)}{\sqrt{2}}$  spin antisymmetric

$$\Psi_{total} = \Psi_{spin} \cdot \Psi_{orbital}$$

total antisym

triplet sym · antisym  $\Rightarrow$  odd  $l$  is sum

antisym · sym  $\Rightarrow$  even  $l$  is sum

singlet

deuterium  $D_2$  spin-1 bosons  $\Rightarrow$  total wavefunction is symmetric

$$\Psi_{\text{total}} = \Psi_{\text{spin}} \cdot \Psi_{\text{orbital}} (-1)^l$$

$$\text{sym} = \begin{cases} \text{sym} & \text{--- sym --- even } l \text{ only} \\ \text{antisym} & \text{--- antisym --- odd } l \text{ only} \end{cases}$$

