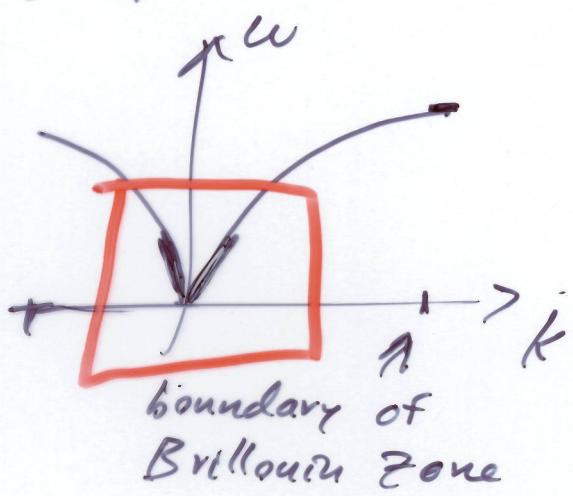


Debye Model

Try to get the T^3 behavior of C_V valid at low temperature

Dispersion Relation (monatomic basis)



Speed of phonons is constant = v

Debye dispersion relation

$$\omega = v k$$

$$\frac{d\omega}{dk} = v$$

Density of states

$$D(\omega) = \frac{\nabla k^2}{2\pi^2} \frac{1}{V_{\text{group}}} = \frac{V\omega^2}{2\pi^2 v^3}$$

Constraint

$$\int_0^{\omega_0} D(\omega) d\omega = \int \frac{dN}{d\omega} d\omega = \int dN = N$$

ω_0 - Debye frequency - cutoff frequency

$$\int_0^{\omega_0} \frac{V\omega^2}{2\pi^2 v^3} d\omega = \frac{V}{2\pi^2 v^3} \cdot \frac{1}{3} \omega^3 \Big|_0^{\omega_0} = \frac{V\omega_0^3}{6\pi^2 v^3} = N$$

$$\omega_0 = \left[\frac{6\pi^2 N}{V} \right]^{1/3}$$

Debye wavenumber $k_D = \frac{\omega_0}{v} = \left[\frac{6\pi^2 N}{V} \right]^{1/3}$

Debye Temperature : $\theta_D = T_D$

$$k_B T_D = \text{energy} = \hbar \omega_0 \Rightarrow T_D = \frac{\hbar \omega_0}{k_B}$$

$$T_D = \frac{\hbar v}{k_B} \left[\frac{6\pi^2 N}{V} \right]^{1/3}$$

$$U = \int D(\omega) \underbrace{\langle n \rangle}_{\text{red}} \underbrace{\hbar \omega}_{\text{blue}} d\omega$$

$$U = \underbrace{\int_0^{\omega_0} \left(\frac{V\omega^2}{2\pi^2 v^3} \right)}_{\text{per polarization}} \underbrace{\frac{1}{e^{\frac{\hbar\omega}{k_B T}} - 1}}_{\text{only temperature dependence}} \underbrace{\hbar \omega d\omega}_{\text{green}}$$

Assume v is the same for all 3 polarizations
1 longitudinal 2 transverse

$$U = \frac{3V\hbar}{2\pi^2 c^3} \int_0^{\omega_0} \frac{\omega^3 d\omega}{e^{\frac{\hbar\omega}{k_B T}} - 1}$$

$$C_V = \left(\frac{\partial U}{\partial T} \right)_V = \frac{3V\hbar}{2\pi^2 c^3} \int_0^{\omega_0} \frac{\omega^3 d\omega}{\left[e^{\frac{\hbar\omega}{k_B T}} - 1 \right]^2} \cdot e^{\frac{\hbar\omega}{k_B T}} \cdot \frac{\hbar\omega}{k_B T} (-1)$$

$$= \frac{3V\hbar^2}{2\pi^2 c^3 k_B T^2} \int_0^{\omega_0} \frac{\omega^4 e^{\frac{\hbar\omega}{k_B T}} d\omega}{\left[e^{\frac{\hbar\omega}{k_B T}} - 1 \right]^2}$$

Low T behavior should be $\propto T^3$ - check

$$U = ? \quad \text{Define } X = \frac{\hbar\omega}{k_B T} \quad , \quad X_D = \frac{\hbar\omega_0}{k_B T} = \frac{T_0}{T}$$

$$= \frac{6\omega_0}{T}$$

$$U = 9Nk_B T \left(\frac{T}{T_0} \right)^3 \int_0^{X_D} \frac{x^3}{e^x - 1} dx$$

for low T $\ll T_0$, replace X_D by ∞

$$U = 9Nk_B T \left(\frac{T}{T_0} \right)^3 \boxed{\int_0^{\infty} \frac{x^3}{e^x - 1} dx} = 6 \zeta(4)$$

$$= \frac{\pi^4}{15}$$

$$C_V = \left(\frac{\partial U}{\partial T} \right)_V = \frac{2}{3T} \left[\frac{3}{S} \frac{Nk_B T^4 \pi^4}{T_D^3} \right]$$

$$= \frac{12\pi^4 N k_B}{S} \left(\frac{T}{T_D} \right)^3 \leftarrow \text{Debye } T^3 \text{ law}$$

Valid for low T , also correct high T
limit - Dulong-Petit law $C_V = 3Nk_B$

Einstein Model

Independent harmonic oscillator

one frequency ω_E

$$D(\omega) = A \delta(\omega - \omega_E)$$



$$\text{Total \# of modes} = \int D(\omega) d\omega = N$$

$$D(\omega) = N \delta(\omega - \omega_E)$$

$$U = \int_0^\infty D(\omega) \langle n \rangle \hbar \omega d\omega = \int_0^\infty N \delta(\omega - \omega_E) \frac{\hbar \omega}{e^{\frac{\hbar \omega}{k_B T}} - 1} d\omega$$

$$= \frac{N \hbar \omega_E}{e^{\frac{\hbar \omega_E}{k_B T}} - 1}$$

$$U = \frac{N \hbar \omega_E}{e^{\frac{\hbar \omega_E}{k_B T}} - 1}$$

$$C_V = \left(\frac{\partial U}{\partial T} \right)_V = \frac{N \hbar \omega_E (-1) e^{\frac{\hbar \omega_E}{k_B T}} \frac{\hbar \omega_E}{k_B T^2} (-1)}{\left[e^{\frac{\hbar \omega_E}{k_B T}} - 1 \right]^2}$$

$$= N k_B \left(\frac{\hbar \omega_E}{k_B T} \right)^2 \frac{e^{\frac{\hbar \omega_E}{k_B T}}}{\left[e^{\frac{\hbar \omega_E}{k_B T}} - 1 \right]^2}$$

Multiply by 3 for 3 polarization

high T limit

$$\lim_{x \rightarrow 0} \frac{x^2 e^x}{[e^x - 1]^2} = 1$$

$$C_V \rightarrow 3 N k_B \quad (\text{Dulong & Petit})$$

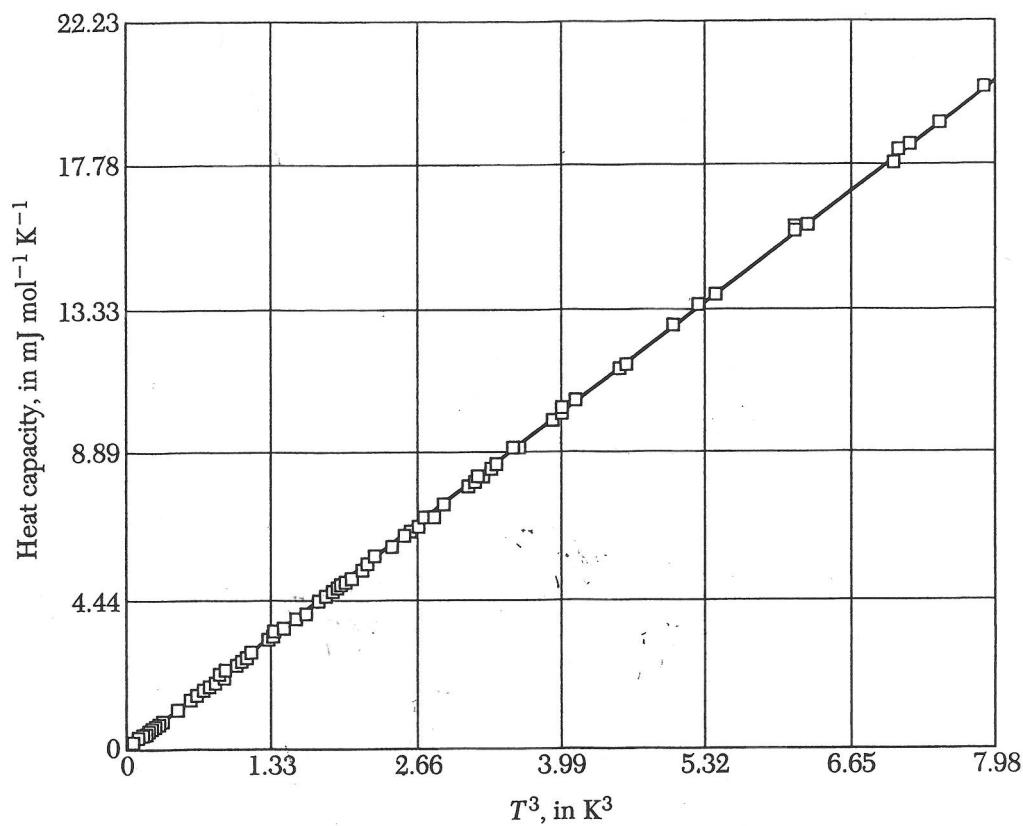


Figure 9 Low temperature heat capacity of solid argon, plotted against T^3 . In this temperature region the experimental results are in excellent agreement with the Debye T^3 law with $\theta = 92.0$ K. (Courtesy of L. Finegold and N. E. Phillips.)