

Bulk Modulus (Elastic Modulus)  $F = -kx$

$$K = B = \frac{\text{stress}}{\text{strain}} = -V \frac{\partial P}{\partial V} \leftarrow \text{pressure}$$

$\uparrow$   
Volume

$$= \frac{-dP \text{ (stress)}}{\left(\frac{dV}{V}\right) \text{ (strain)}}$$

Thermodynamics:  $T dS = dU + P dV$

$\circ$   $T \approx 0K$

*- some texts*

$$P = - \frac{\partial U}{\partial V} \Rightarrow - \frac{\partial P}{\partial V} = \frac{\partial^2 U}{\partial V^2} \Rightarrow B = V \frac{\partial^2 U}{\partial V^2}$$

$$\frac{\partial U}{\partial V} = \frac{\partial U}{\partial R} \frac{\partial R}{\partial V}$$

$$\left. \frac{\partial^2 U}{\partial V^2} \right|_{R=R_0} = \left[ \frac{\partial^2 U}{\partial R^2} \left(\frac{\partial R}{\partial V}\right)^2 + \frac{\partial U}{\partial R} \frac{\partial^2 R}{\partial V^2} \right] \Big|_{R=R_0}$$

$\uparrow$

$\rightarrow 0$  when  $R=R_0$

Ionic crystal

95

$$U(R) = N \left( z\lambda e^{-\frac{R}{\rho}} - \frac{\alpha q^2}{R} \right)$$

$$\frac{dU}{dR} = N \left( -\frac{z\lambda}{\rho} e^{-\frac{R}{\rho}} + \frac{\alpha q^2}{R^2} \right)$$

$$\frac{d^2U}{dR^2} = N \left( \frac{z\lambda}{\rho^2} e^{-\frac{R}{\rho}} - \frac{2\alpha q^2}{R^3} \right)$$

$$\left. \frac{d^2U}{dR^2} \right|_{R=R_0} = N \left( \frac{z\lambda}{\rho^2} e^{-\frac{R_0}{\rho}} - \frac{2\alpha q^2}{R_0^3} \right)$$

$$\frac{\rho \alpha q^2}{z\lambda R_0^2}$$

$$= N \left[ \frac{z\lambda}{\rho^2} \left( \frac{\rho \alpha q^2}{z\lambda R_0^2} \right) - \frac{2\alpha q^2}{R_0^3} \right]$$

$$= N \frac{\alpha q^2}{R_0^3} \left( \frac{R_0}{\rho} - 2 \right)$$

$$V = \frac{Na^3}{4}$$

$$a = 2R$$

$$\Rightarrow V = 2NR^3$$

fcc

$$\frac{\partial V}{\partial R} = 6NR^2 \Rightarrow \left( \frac{\partial R}{\partial V} \right)^2 = \frac{1}{36N^2R^4}$$

$$\left. \frac{B}{R_0} \right|_R = V \left. \frac{\partial^2 U}{\partial V^2} \right|_R = V \left. \frac{\partial^2 U}{\partial R^2} \left( \frac{\partial R}{\partial V} \right)^2 \right|_R$$

$$= (2NR_0^3) \left[ \frac{N\alpha q^2}{R_0^3} \left( \frac{R_0}{s} - 2 \right) \right] \left( \frac{1}{36N^2 R_0^4} \right)$$

$$= \frac{\alpha q^2}{18R_0^4} \left( \frac{R_0}{s} - 2 \right)$$

Zn S

diamond

ionic or covalent

Zn<sup>++</sup> S<sup>--</sup>

~~Zn S~~