

SOLID-STATE PHYSICS: CONDUCTION

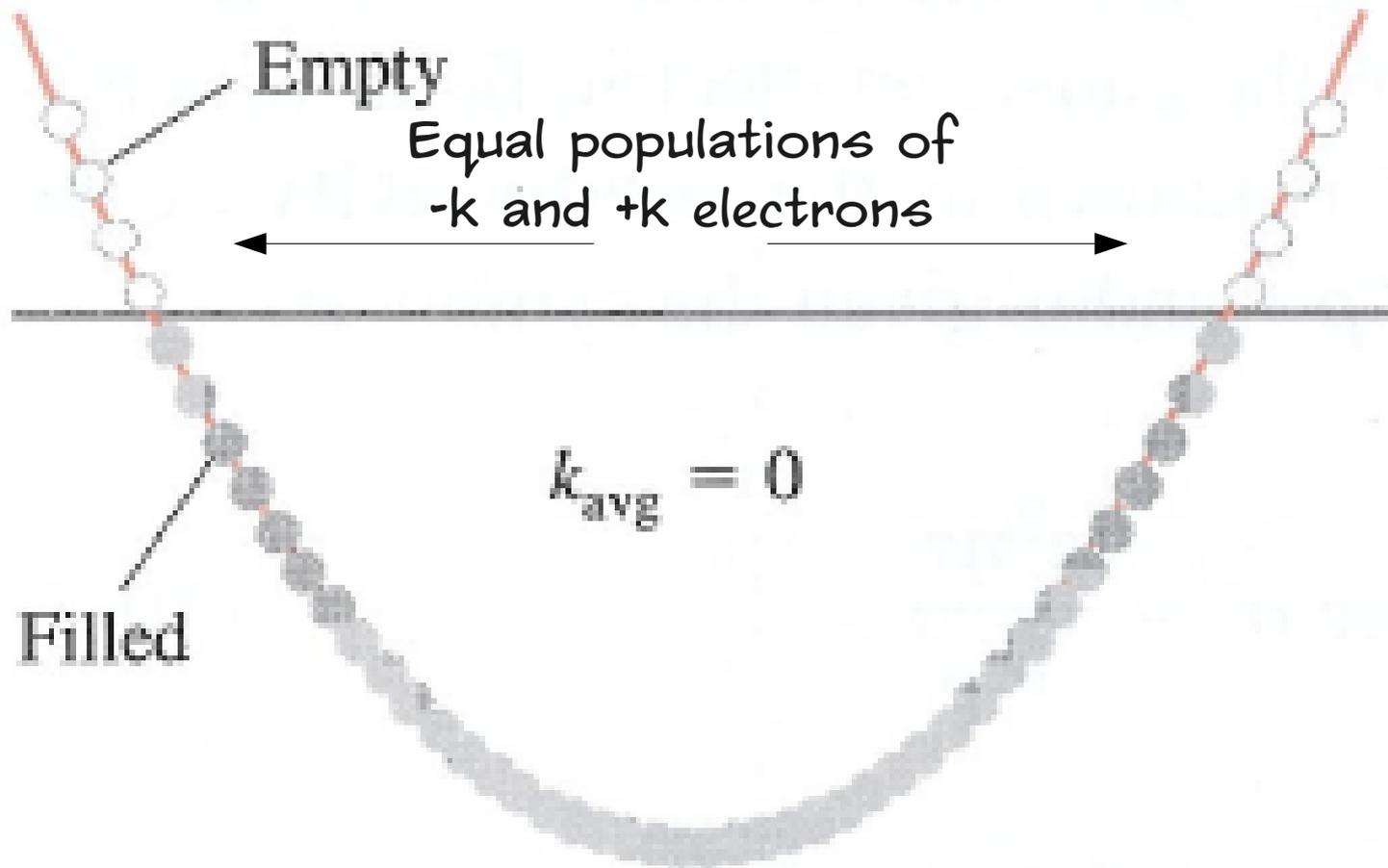
Prof. Stephen Sekula
(3/30/2010)

Supplementary Material for
PHY 3305 (Modern Physics)
Harris, Ch. 10.6-10.8

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SOLID, NO ELECTRIC FIELD APPLIED

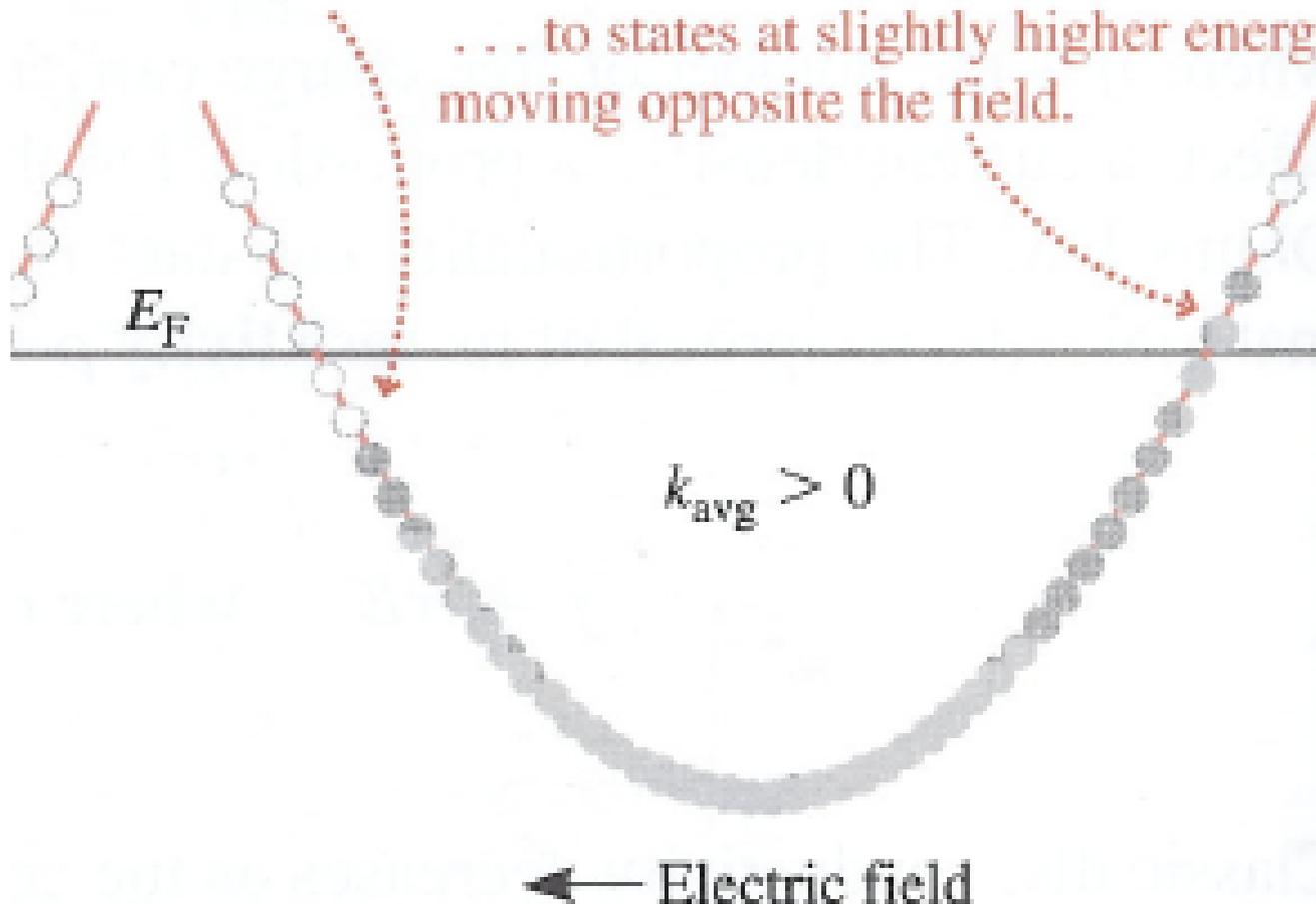


Electric field = 0

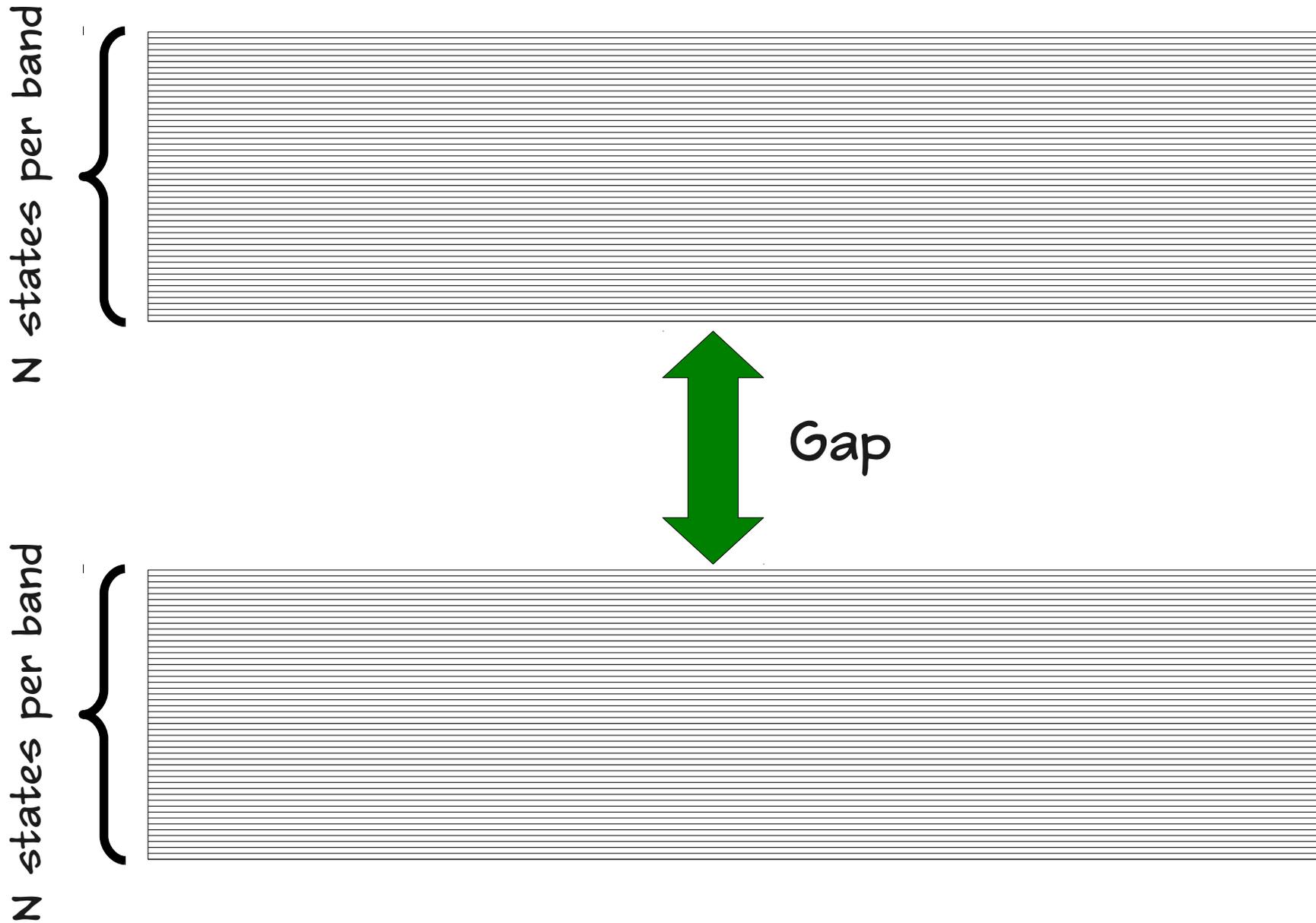
SOLID, WITH ELECTRIC FIELD APPLIED

The net effect of a field is to shift some electrons moving in the field's direction . . .

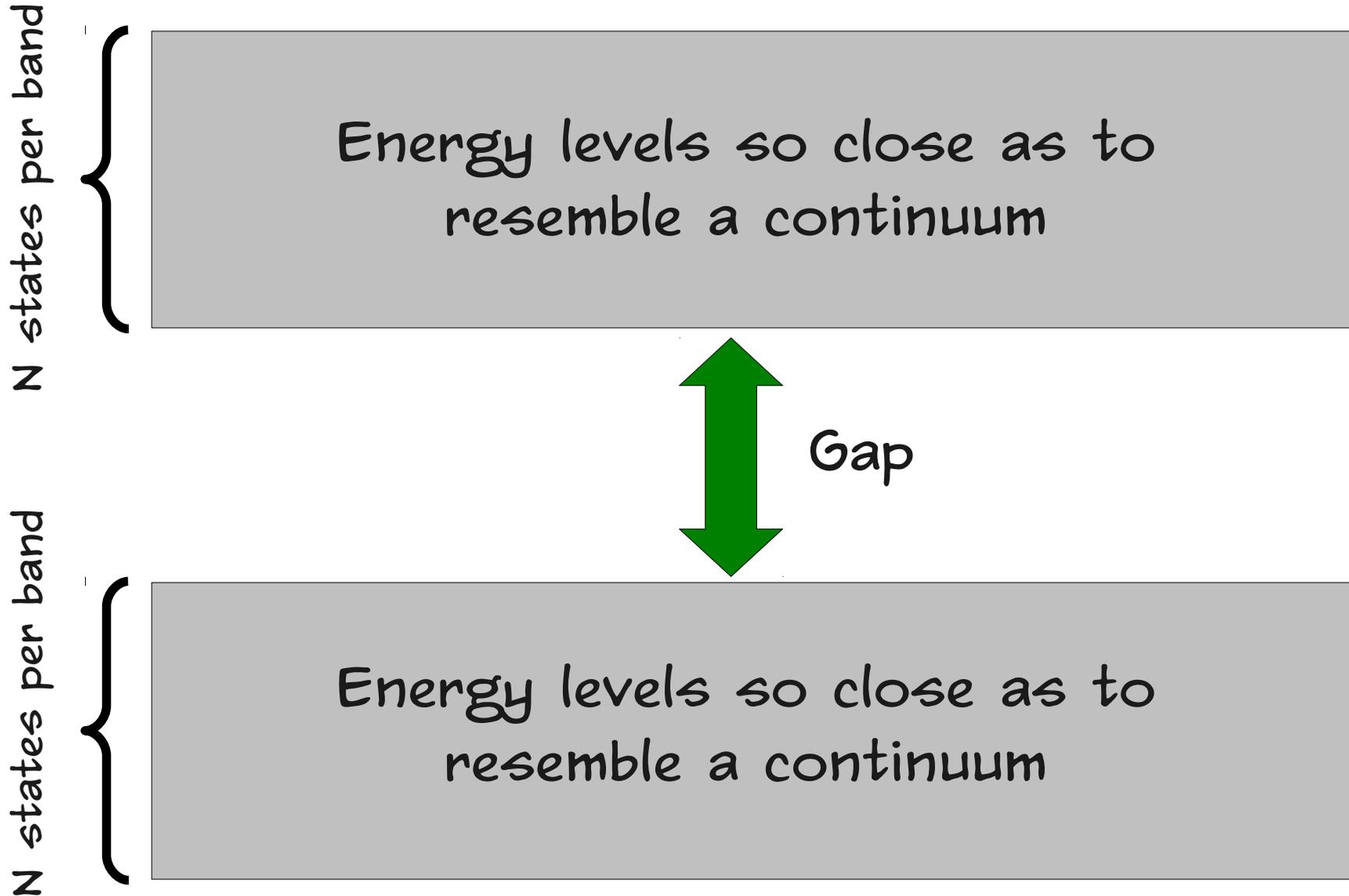
. . . to states at slightly higher energy moving opposite the field.



A PICTURE: BANDS/GAPS



A PICTURE: BANDS/GAPS



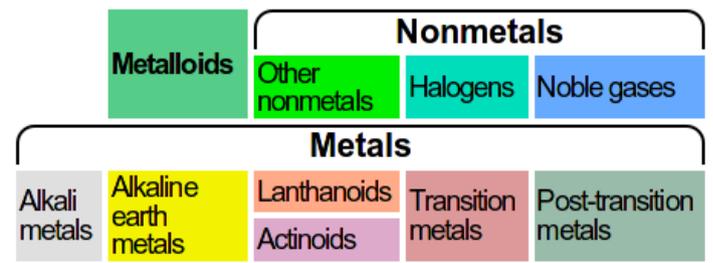
QUESTIONS

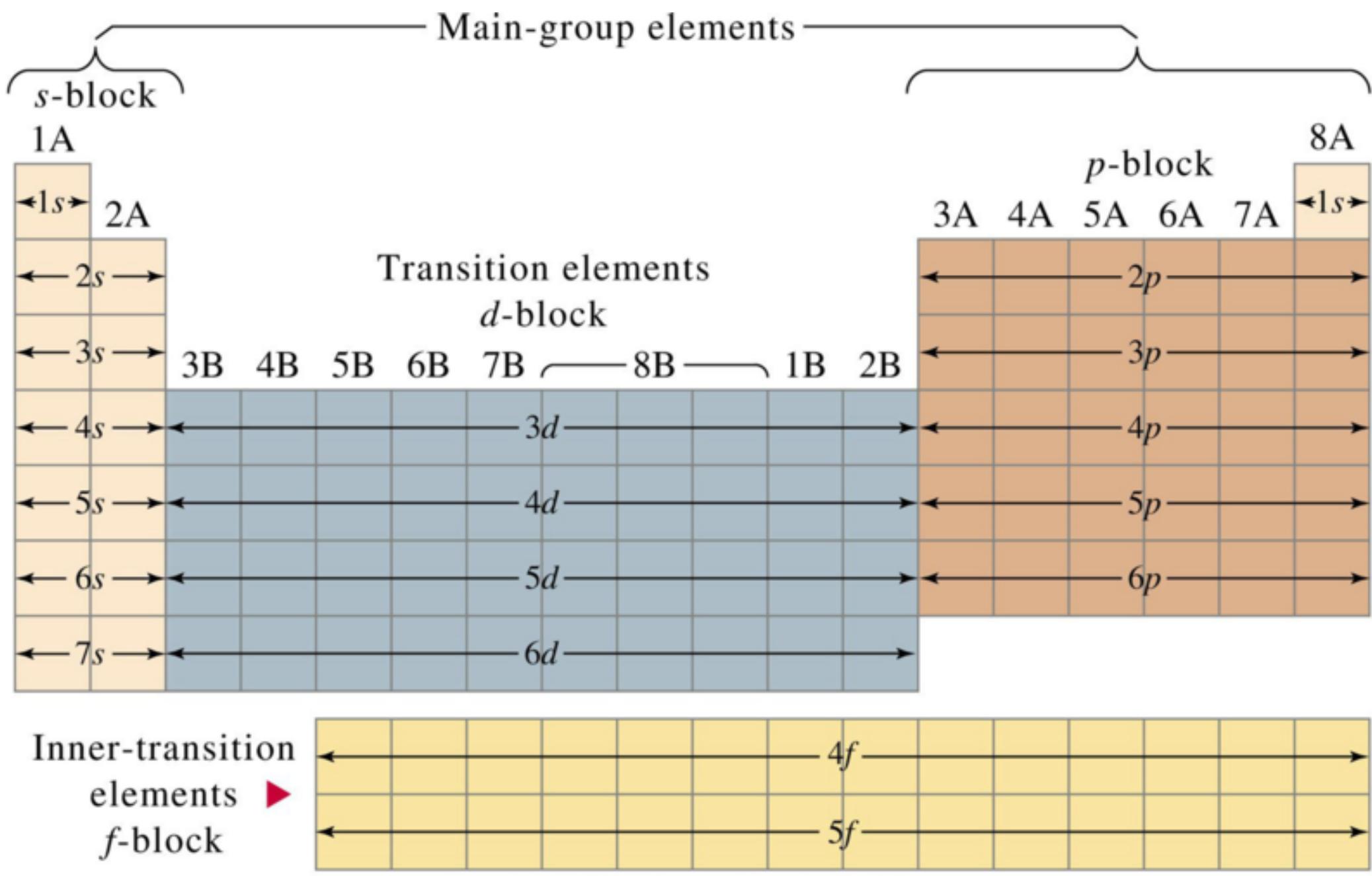
- How many conduction electrons can each band hold?
- Which is a better conductor: Lithium or Beryllium?
 - Lithium has 3 electrons
 - Beryllium has 4 electrons

Periodic Table of Elements

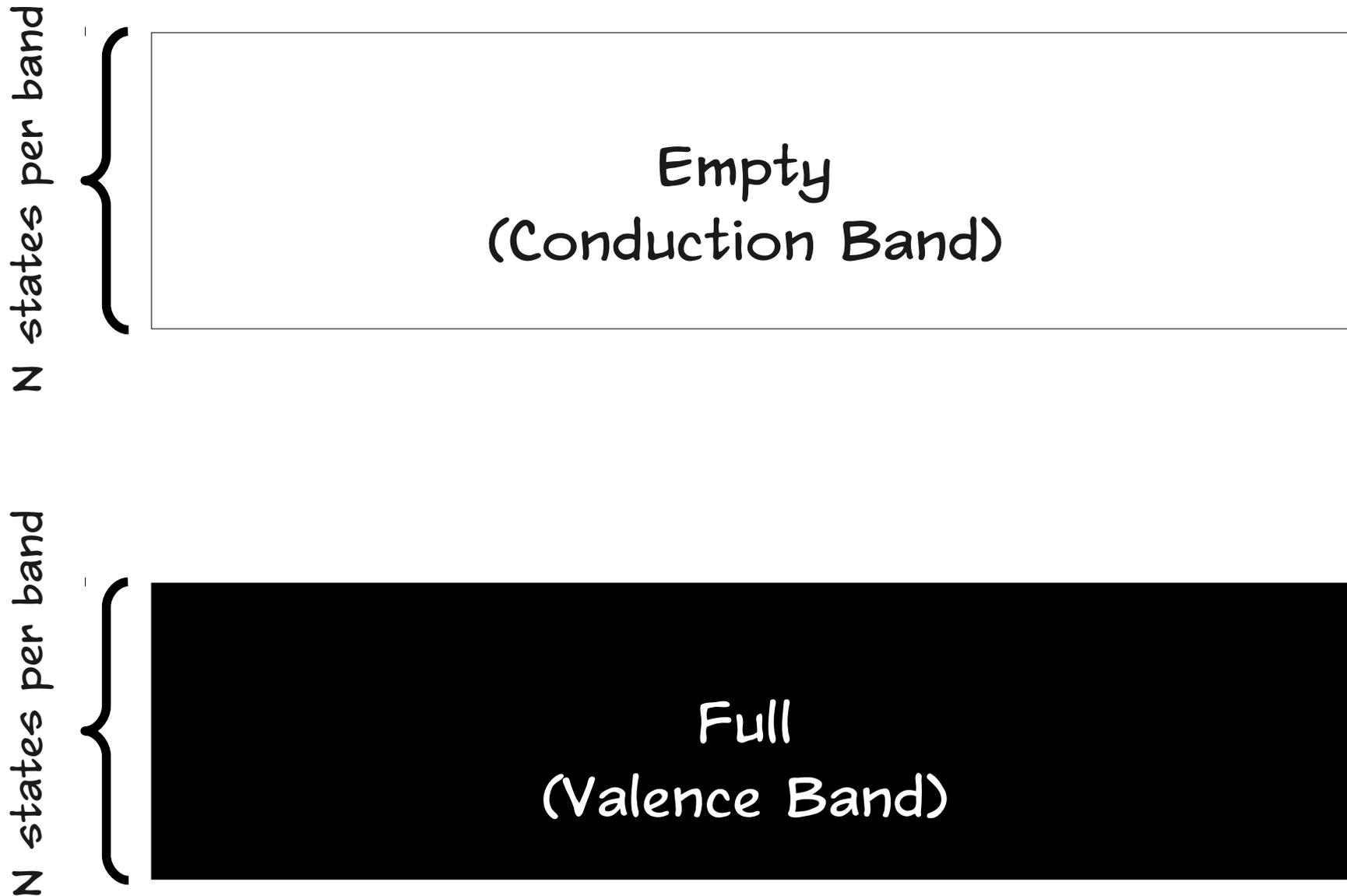
 Weight
 Names
 Electrons
 Wide

	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18												
1	H Hydrogen 1.00794																	2	He Helium 4.002602											
2	Li Lithium 6.941	Be Beryllium 9.012182																	3	Li Lithium 6.941	4	Be Beryllium 9.012182								
3	Na Sodium 22.98976928	Mg Magnesium 24.3050																	5	B Boron 10.811	6	C Carbon 12.0107	7	N Nitrogen 14.0067	8	O Oxygen 15.9994	9	F Fluorine 18.9984032	10	Ne Neon 20.1797
4	K Potassium 39.0983	Ca Calcium 40.078	Sc Scandium 44.955912	Ti Titanium 47.867	V Vanadium 50.9415	Cr Chromium 51.9961	Mn Manganese 54.938045	Fe Iron 55.845	Co Cobalt 58.933195	Ni Nickel 58.6934	Cu Copper 63.546	Zn Zinc 65.38	Ga Gallium 69.723	Ge Germanium 72.64	As Arsenic 74.92160	Se Selenium 78.96	Br Bromine 79.904	Kr Krypton 83.798												
5	Rb Rubidium 85.4678	Sr Strontium 87.62	Y Yttrium 88.90585	Zr Zirconium 91.224	Nb Niobium 92.90638	Mo Molybdenum 95.96	Tc Technetium (98)	Ru Ruthenium 101.07	Rh Rhodium 102.90550	Pd Palladium 106.42	Ag Silver 107.8682	Cd Cadmium 112.411	In Indium 114.818	Sn Tin 118.710	Sb Antimony 121.760	Te Tellurium 127.60	I Iodine 126.90447	Xe Xenon 131.293												
6	Cs Caesium 132.90545196	Ba Barium 137.327	57-71	Hf Hafnium 178.49	Ta Tantalum 180.94788	W Tungsten 183.84	Re Rhenium 186.207	Os Osmium 190.23	Ir Iridium 192.217	Pt Platinum 195.084	Au Gold 196.966569	Hg Mercury 200.59	Tl Thallium 204.3833	Pb Lead 207.2	Bi Bismuth 208.98040	Po Polonium (209)	At Astatine (210)	Rn Radon (222)												
7	Fr Francium (223)	Ra Radium (226)	89-103	Rf Rutherfordium (261)	Db Dubnium (268)	Sg Seaborgium (271)	Bh Bohrium (272)	Hs Hassium (270)	Mt Meitnerium (276)	Ds Darmstadtium (281)	Rg Roentgenium (280)	Cn Copernicium (285)	Uut Ununtrium (284)	Uuq Ununquadium (289)	Uup Ununpentium (288)	Uuh Ununhexium (293)	Uus Ununseptium (294)	Uuo Ununoctium (294)												

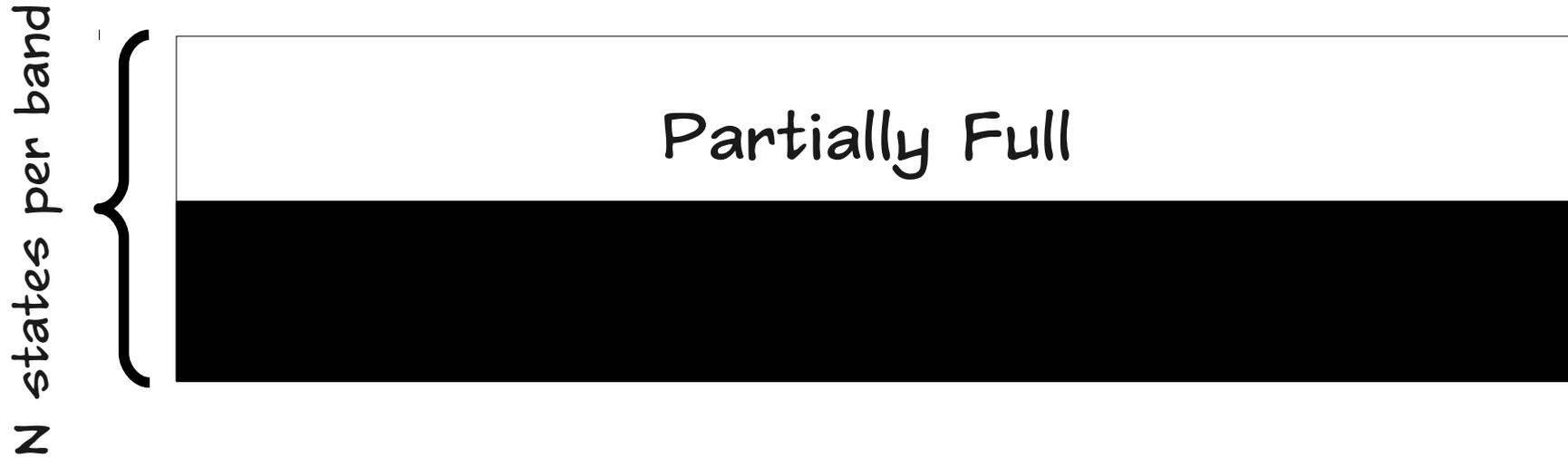




INSULATOR ($T=0$)



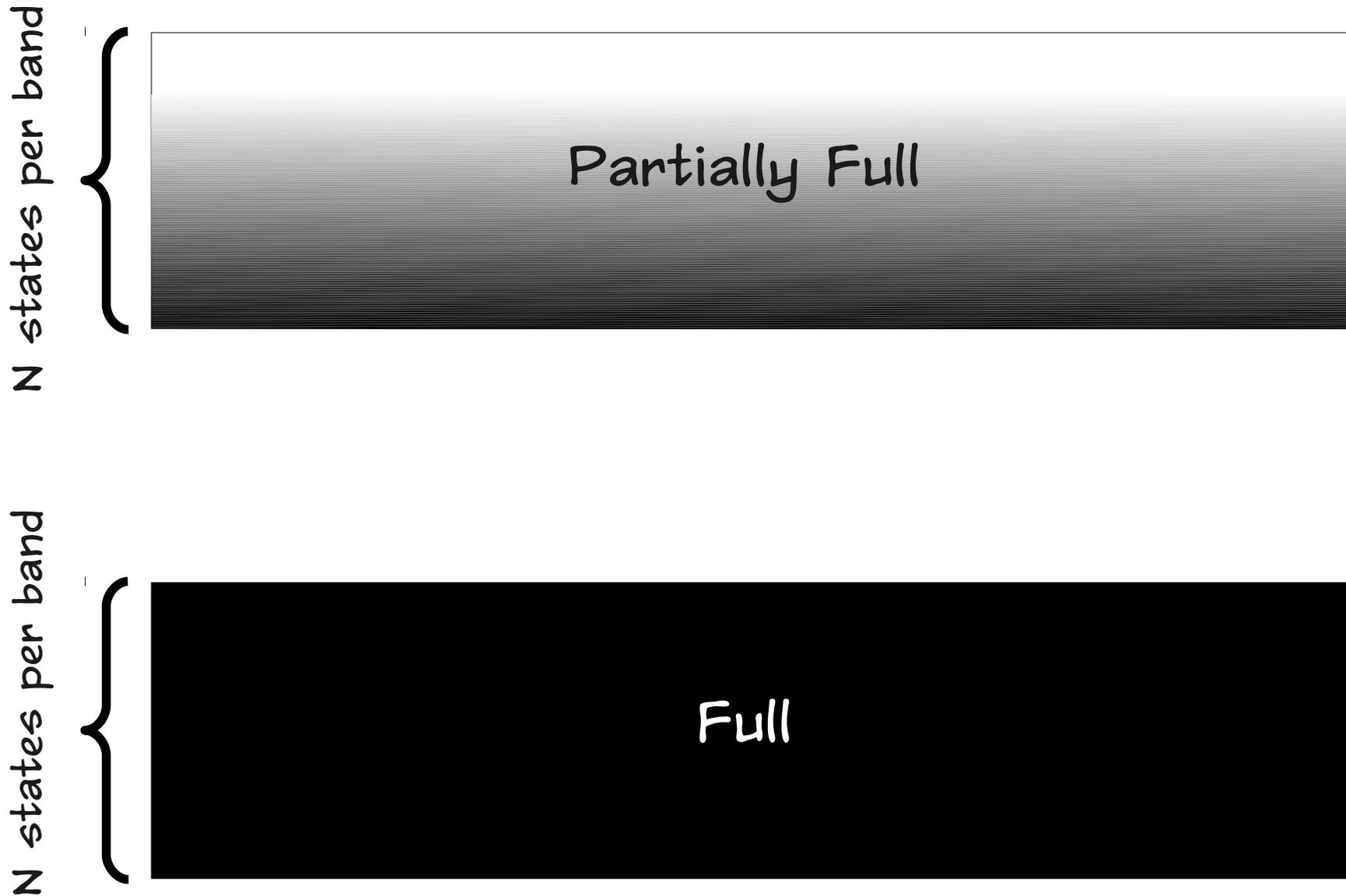
CONDUCTOR ($T=0$)



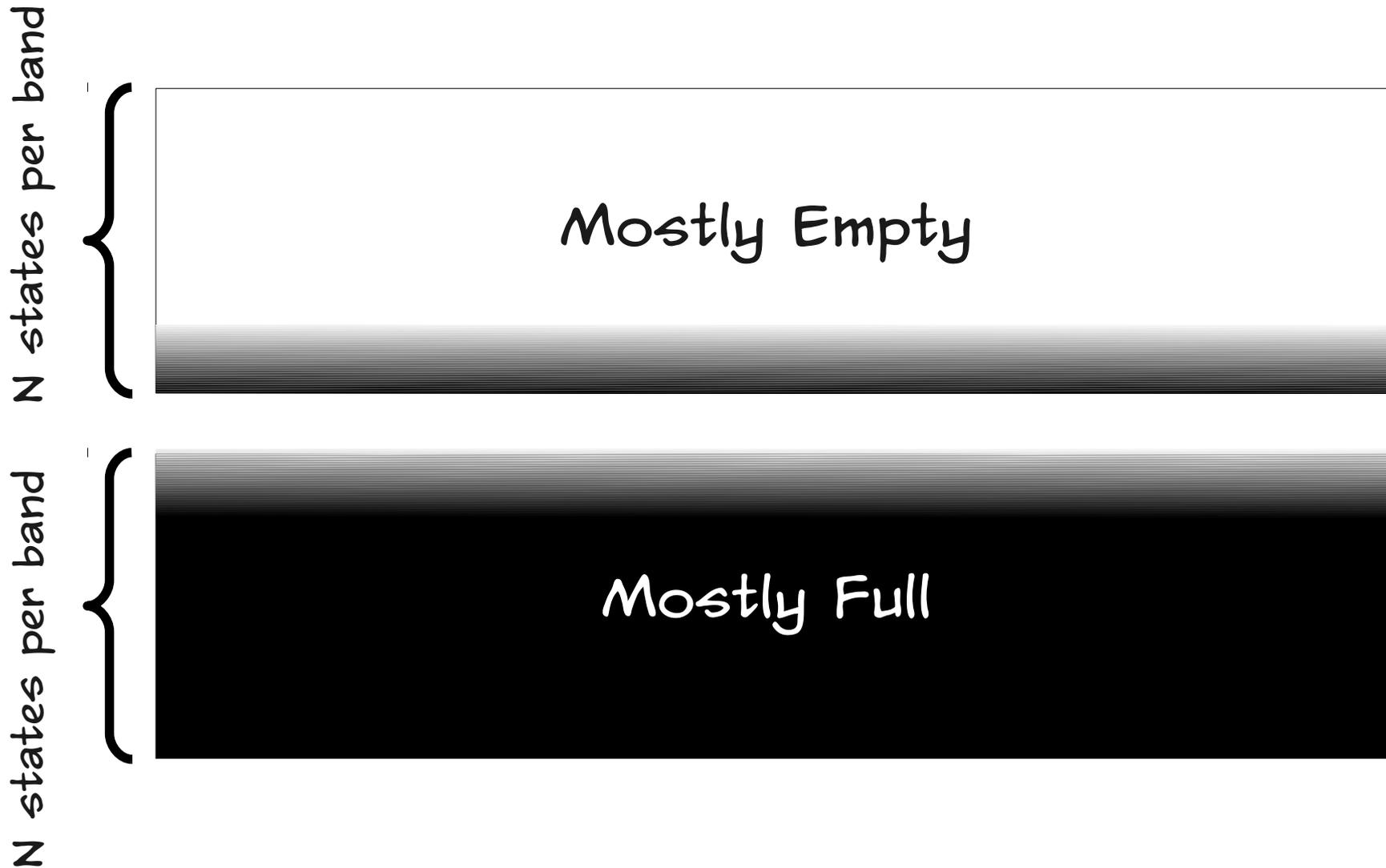
SEMI-CONDUCTOR ($T=0$)



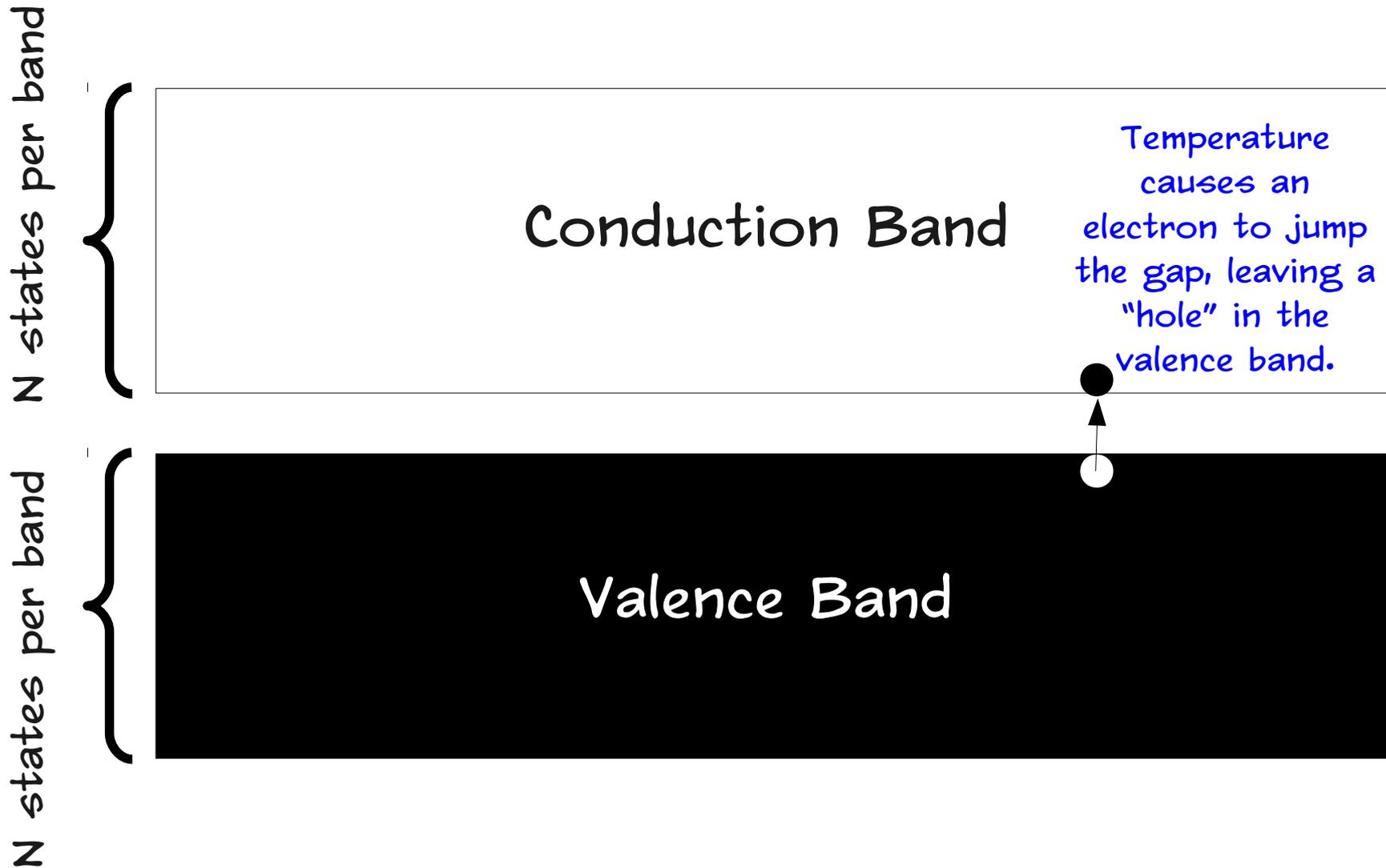
CONDUCTOR ($T > 0$)



SEMI-CONDUCTOR ($T > 0$)

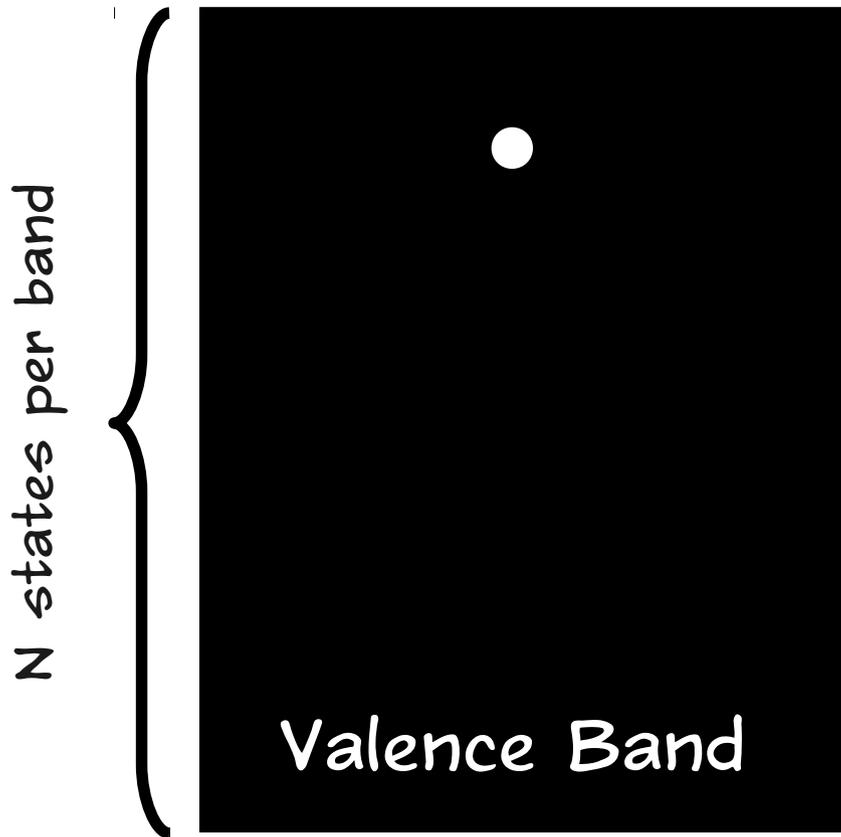


HOLES ($T > 0$)

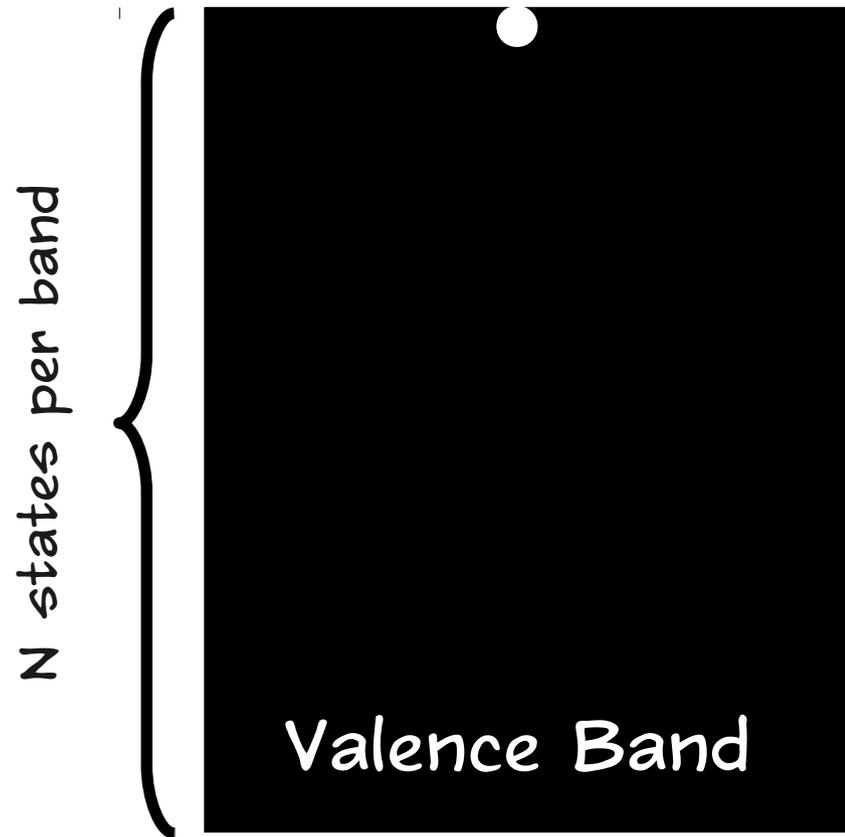


HOLES FLOAT

Total Energy E_1

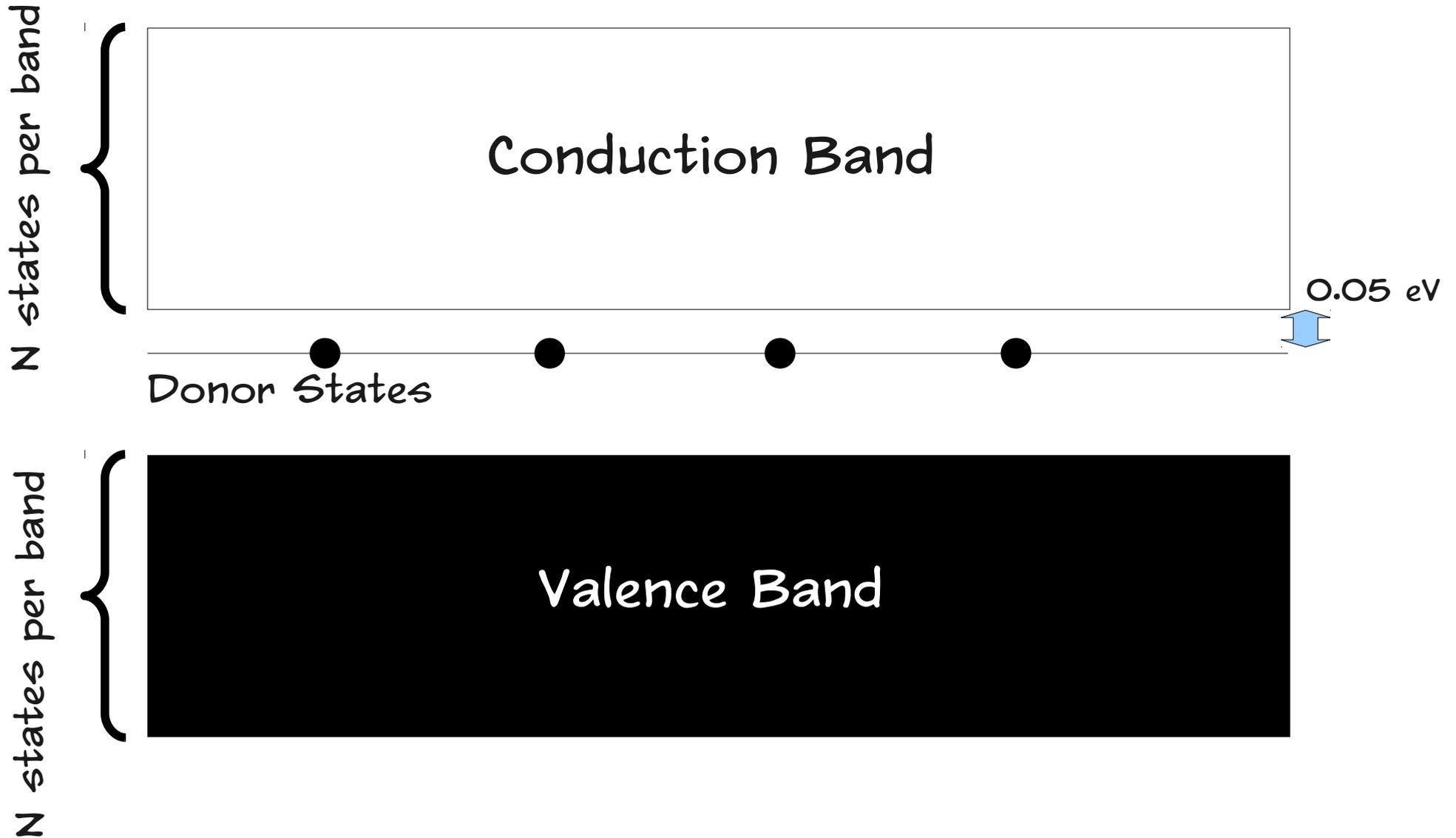


Total Energy E_2

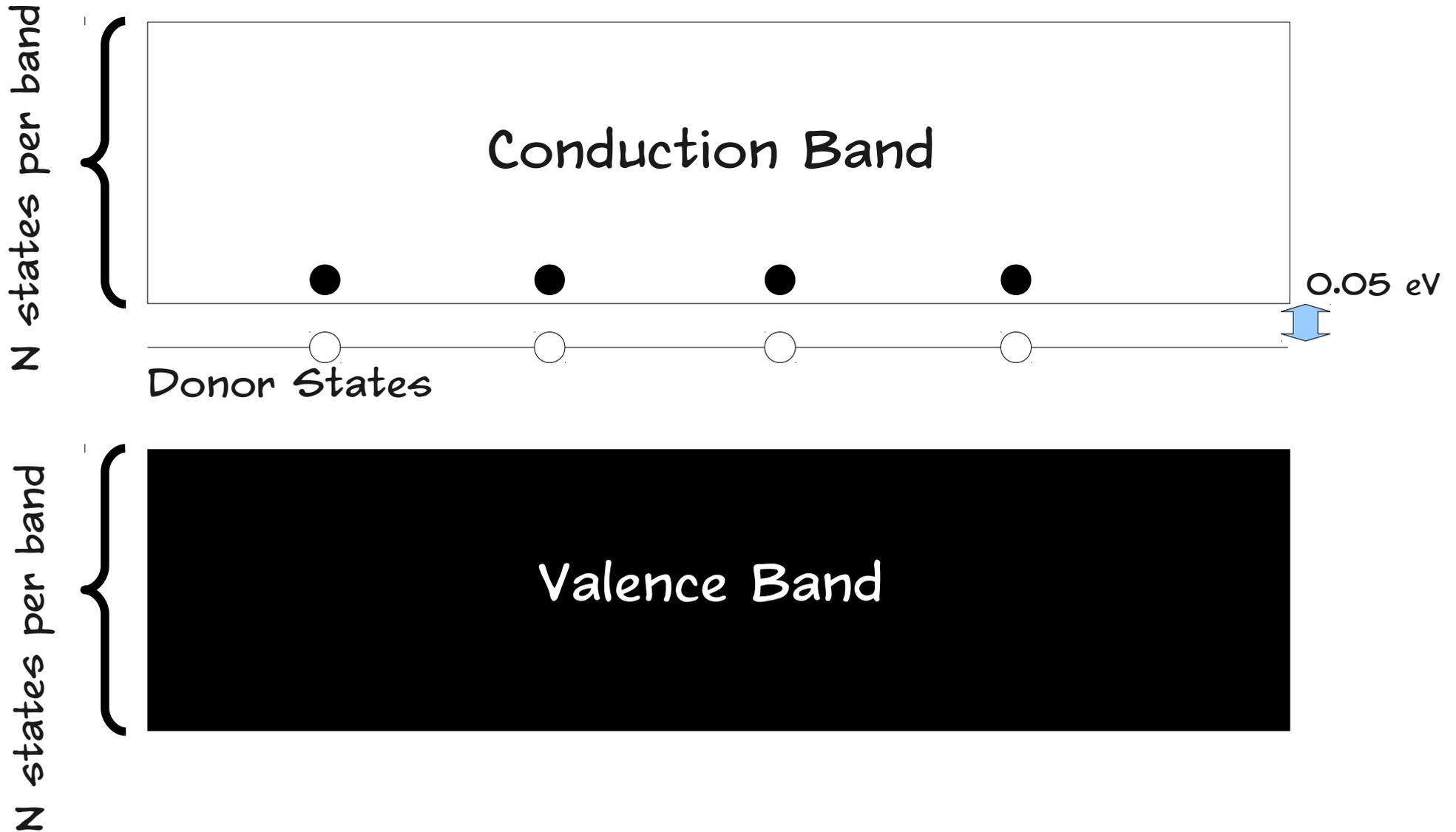


$$E_2 < E_1$$

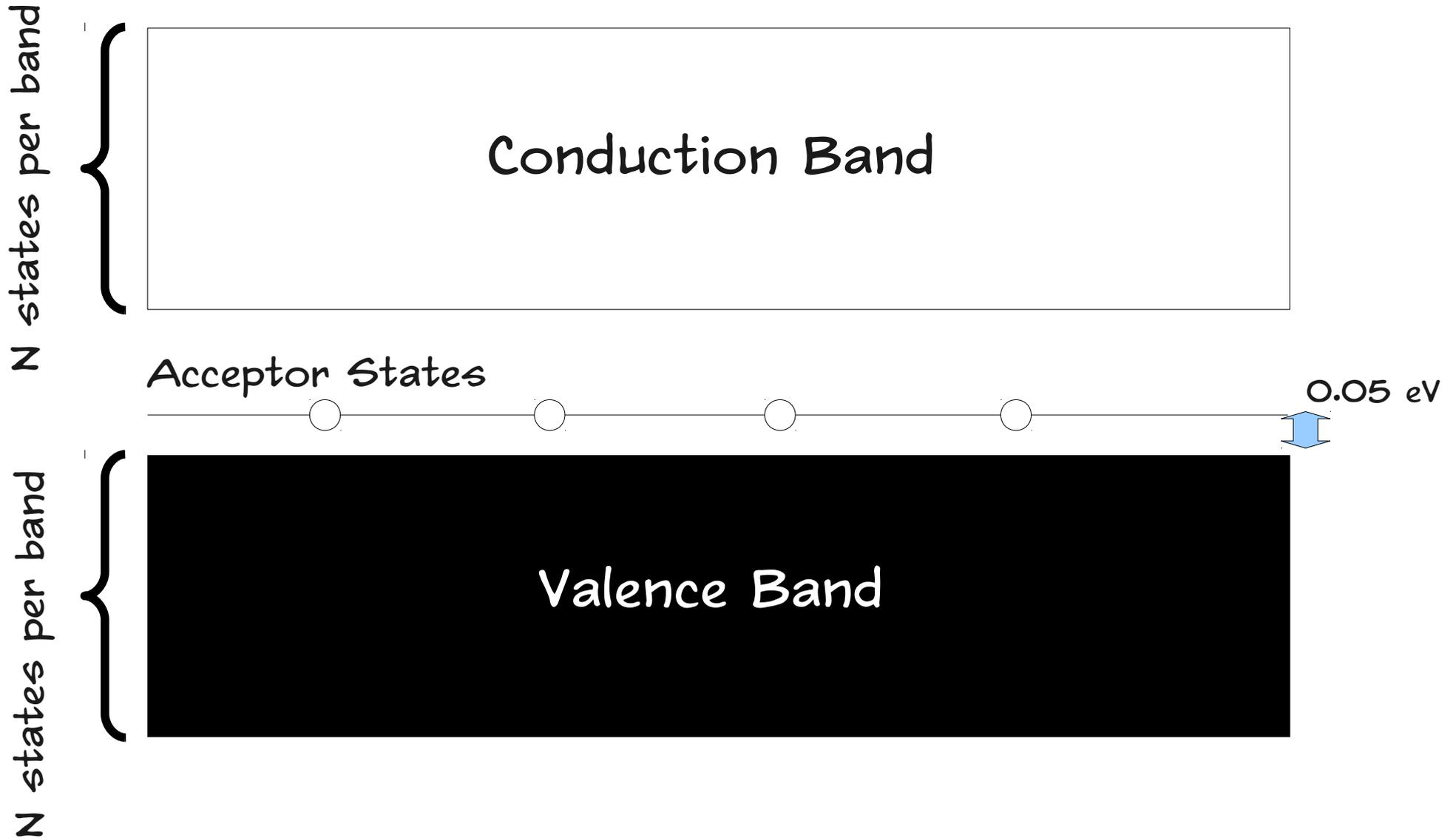
N-TYPE EXTRINSIC SEMICONDUCTOR (T=0)



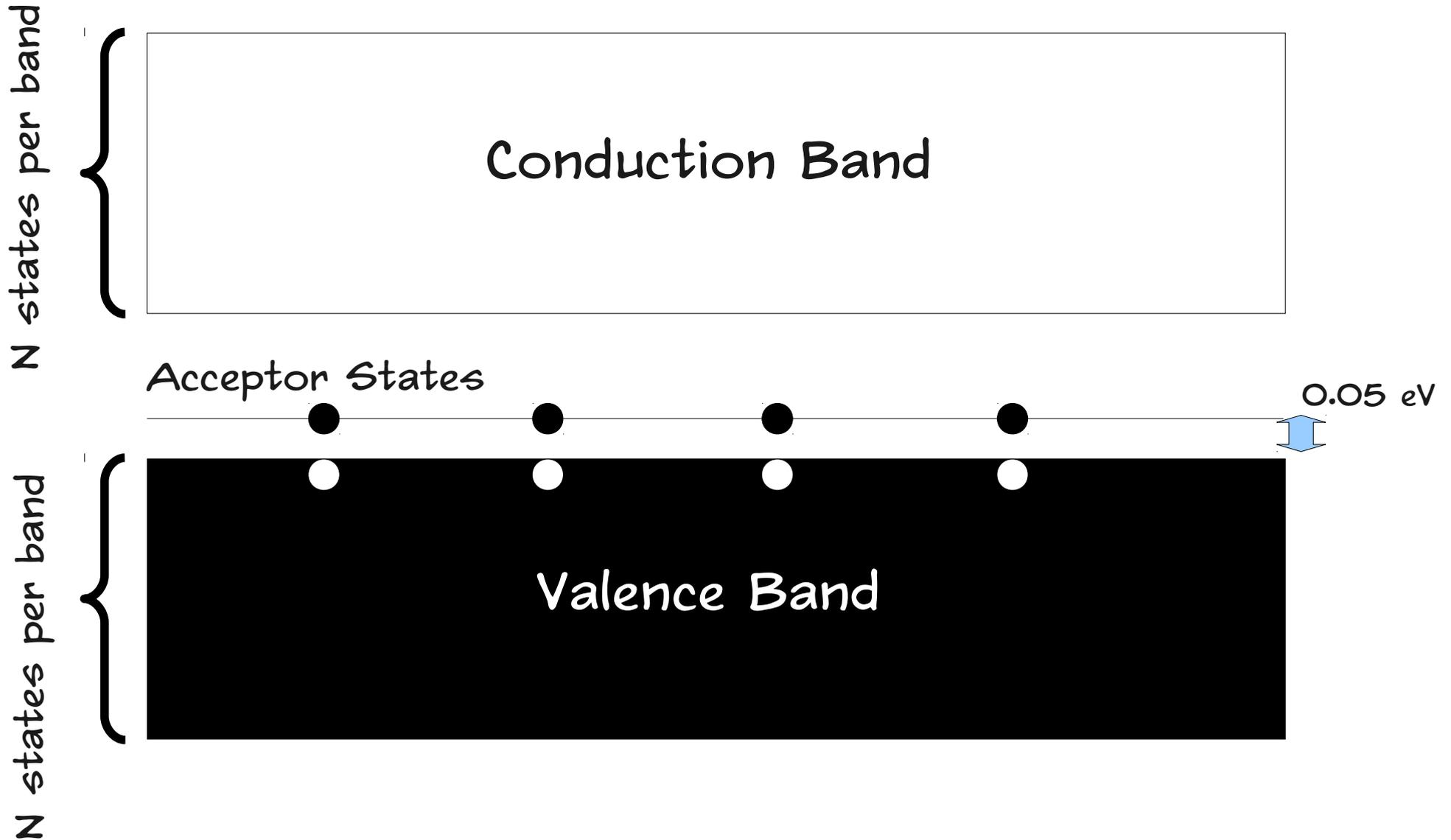
N-TYPE EXTRINSIC SEMICONDUCTOR ($T > 0$)



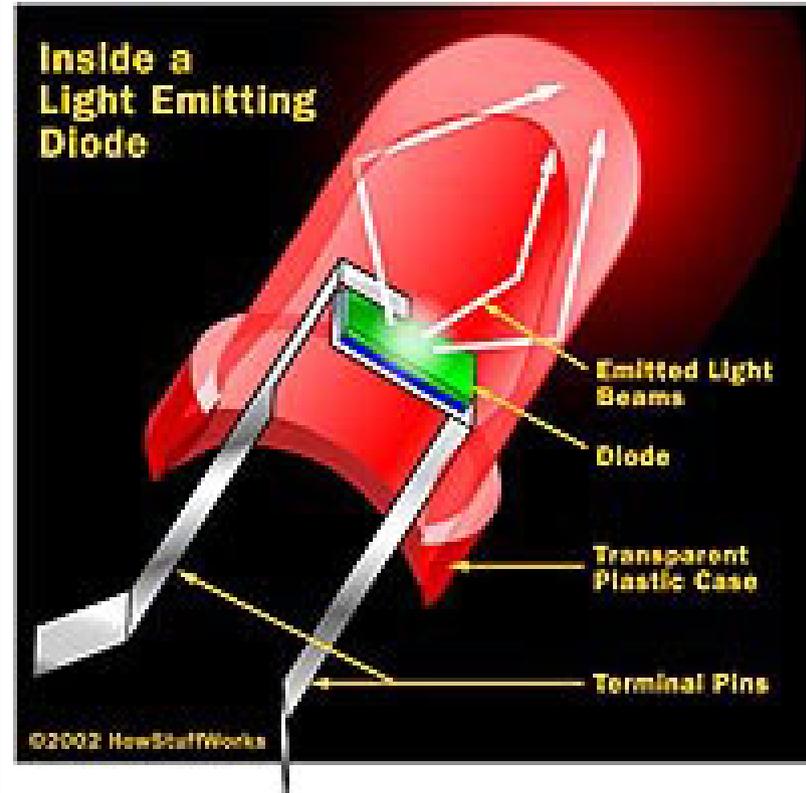
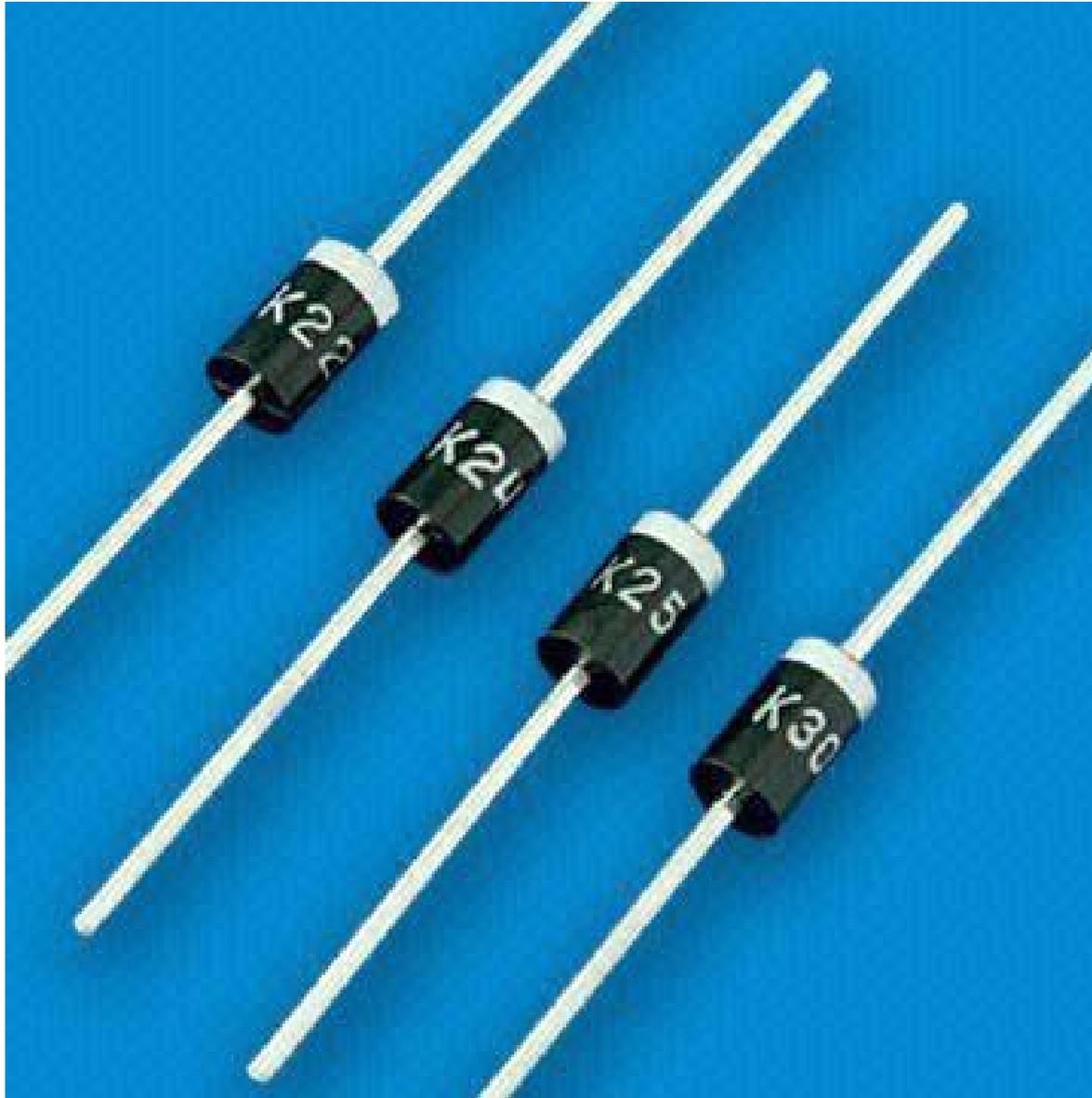
P-TYPE EXTRINSIC SEMICONDUCTOR (T=0)

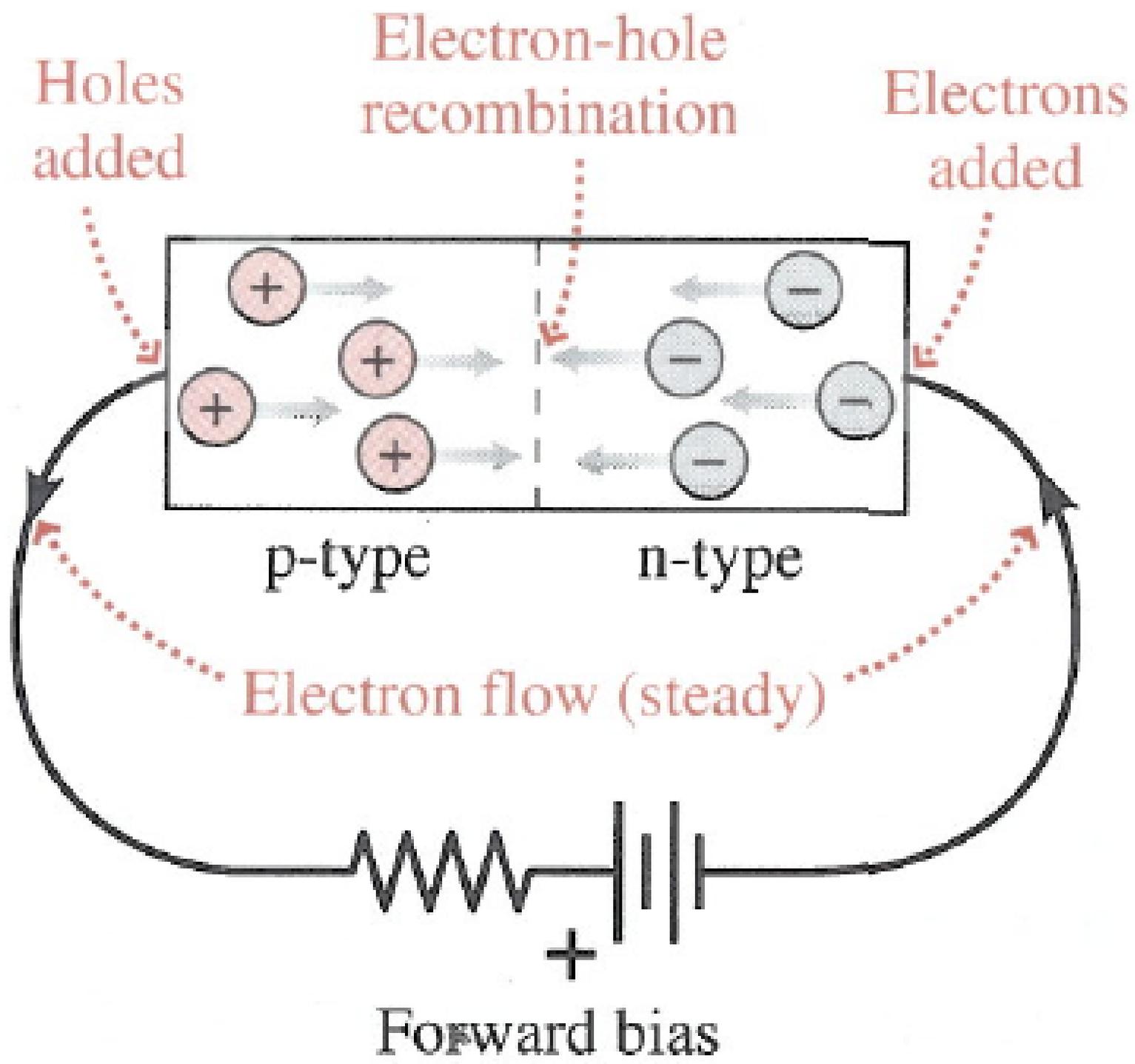


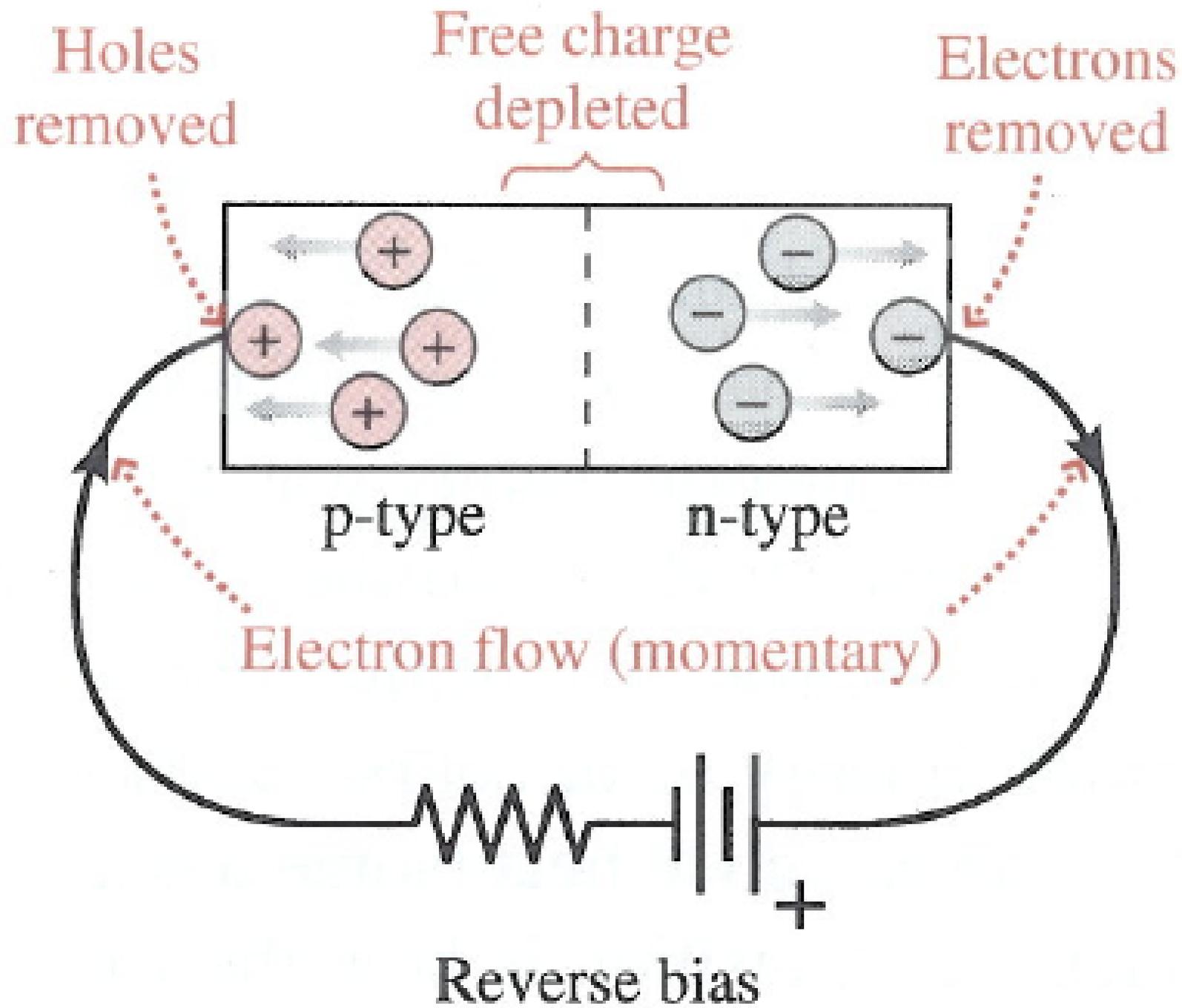
P-TYPE EXTRINSIC SEMICONDUCTOR ($T > 0$)



THE DIODE

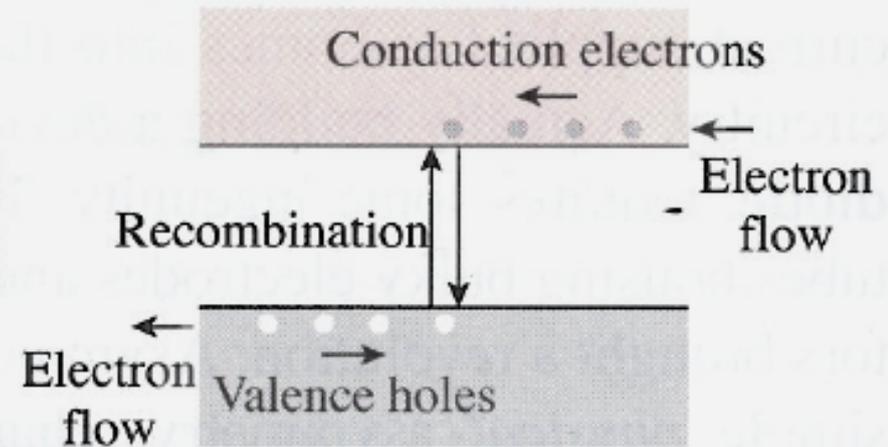
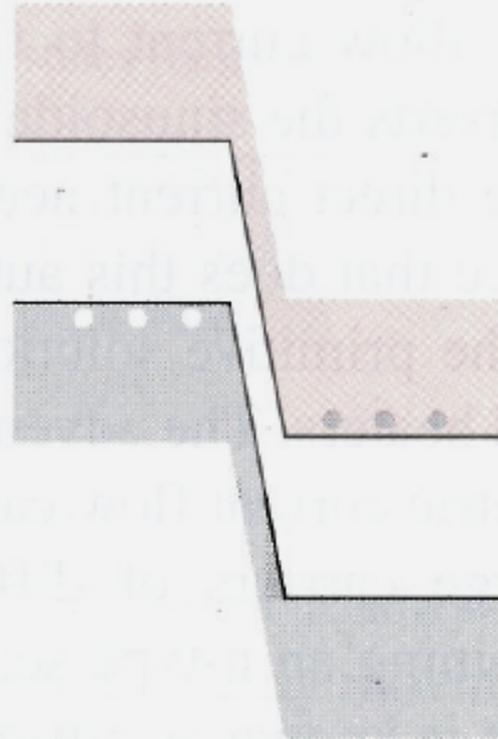
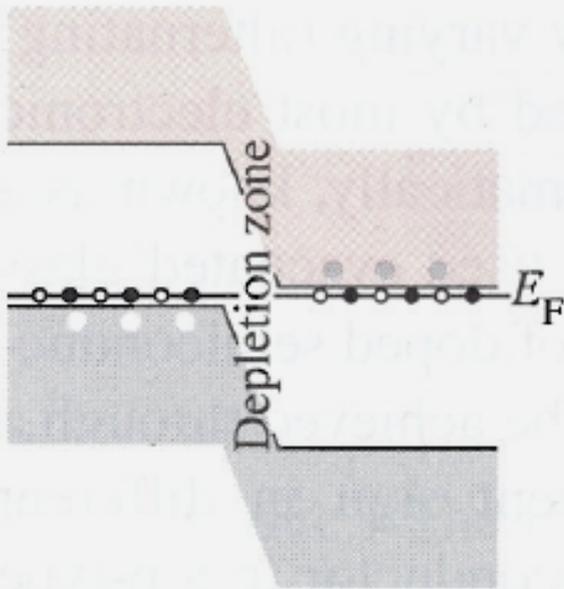
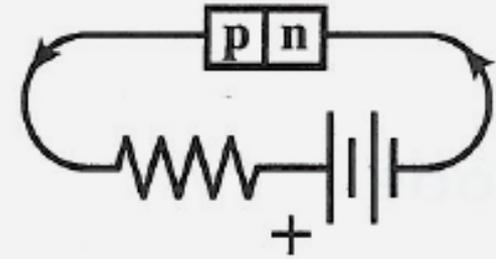
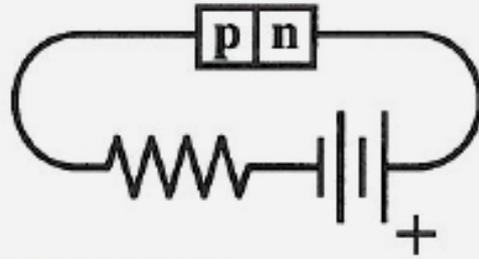






ENERGY BAND PICTURE

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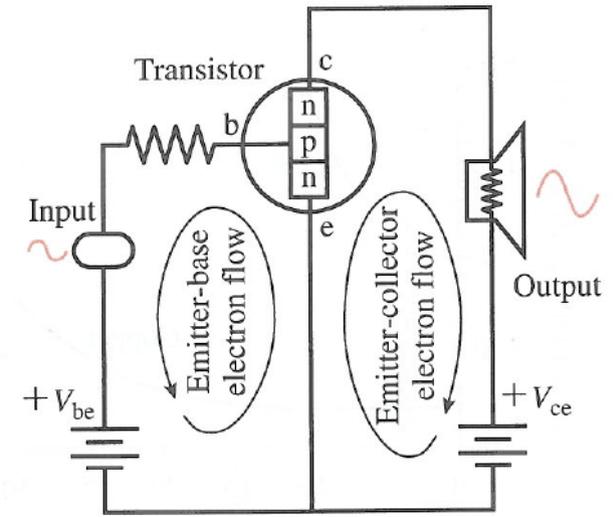
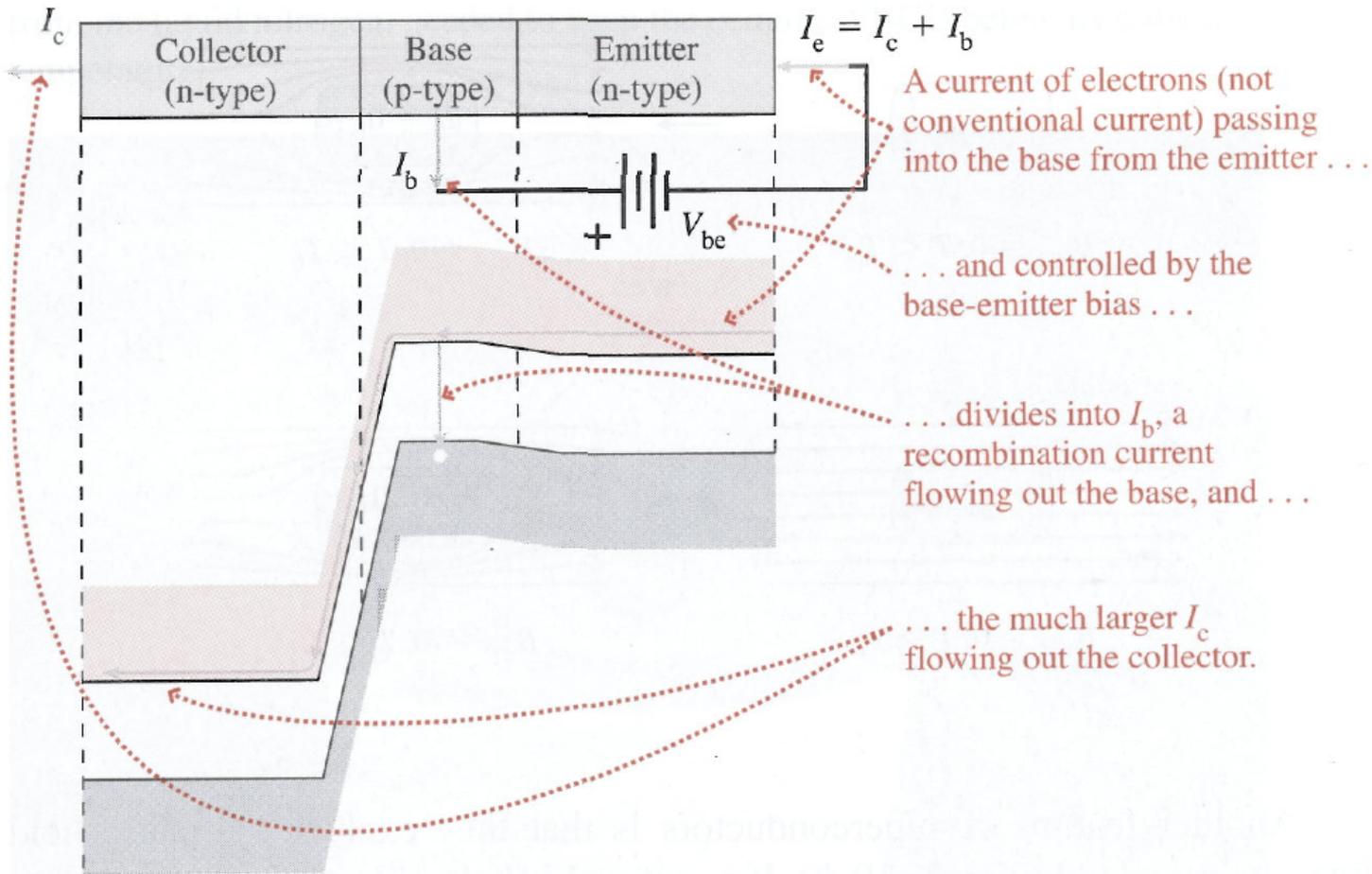


(a) Unbiased

(b) Reverse bias

(c) Forward bias

TRANSISTOR



NEXT TIME

- Superconductivity
- Nuclear Physics
- Good presentation style
- Reading for next time: Harris Ch. 10.9, 11.1-11.2