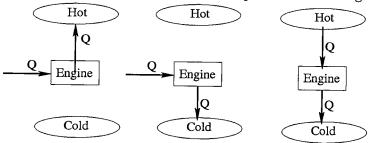
Statistical Mechanics and Thermodynamics August 2016

You may refer to a statistical mechanics text of your choice, but no other notes or references.

- 1. Consider a system A with two spin-1 particles at fixed sites each having a magnetic moment μ . The system is in a magnetic field H pointing along the z direction, is thermally isolated, in equilibrium, and has a total energy of exactly zero. (The only contribution to energy is due to the magnetic moment coupled to the magnetic field. You may ignore kinetic energy.)
 - (1) Determine the probability that the first particle has $s_z = +1\hbar$.
 - (2) This system is brought into thermal contact with a second system B with a single spin-1 particle which originally had $s_z = +1\hbar$ in the same magnetic field. After reaching equilibrium, what is the new probability that the first particle in A has $s_z = +1\hbar$?
 - (3) What are the new average energies E_A and E_B of A and B, and the heat Q_A absorbed by A in part (2)?
- 2. True or false? If false, give a counterexample.
 - (a) _____ All reversible processes are isentropic.
 - (b) _____ All processes with $\Delta Q = 0$ are reversible.
 - (c) _____ All processes with $\Delta Q = 0$ are isentropic.

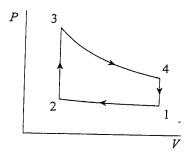
The figures are used in the next three questions about engines and heat reservoirs.



- (d) _____ It is possible to build a machine on which one can perform work and dump all of that energy into the hot reservoir without using the cold reservoir at all.
- (e) _____ It is possible to build a machine on which one can perform work and dump all of that energy into the cold reservoir without using the hot reservoir at all.
- (f) _____ It is possible for heat to flow from the hot reservoir to the cold reservoir with no work at all being done.
- (g) _____ For a system of non-interacting particles (though possibly interacting separately with an external field), the internal energy is always proportional to the absolute temperature.
- (h) _____ If the average velocity of a gas molecule is zero, then the average speed is also zero.

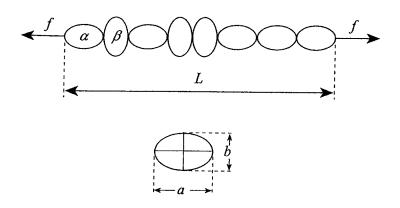
Answer two of the following three problems.

3. The cycle of a highly idealized gasoline engine can be approximated by the Otto cycle, illustrated. The paths $1 \to 2$ and $3 \to 4$ are adiabatic compression and expansion, respectively; $2 \to 3$ and $4 \to 1$ are constant-volume processes. Treat the working medium as an ideal gas with constant $\gamma = c_p/c_v$.



- (a) Compute the efficiency of this cycle for $\gamma=1.4$ and compression ratio $r=V_i/V_f=10$.
- (b) Calculate the work done on the gas in the compression process $1 \to 2$, assuming initial volume $V_i = 2L$ (liters) and $p_i = 1$ atm.
- 4. A brick of mass M sits on the floor, in thermal equilibrium with its surroundings which are at temperature T and act as a reservoir.
 - (a) Using entropy considerations (for example, the relation between heat and change in entropy), give the probability of finding the brick at a height between L and 2L the next time you look. You may treat the brick as a point mass; that is, ignore its own internal energy and contribution to entropy.
 - (b) Use the Boltzmann distribution for the brick to give the probability of finding it at a particular position and momentum. Give the partition function for the brick and use it to compute its average height.

5. Consider a one-dimensional chain consisting of N molecules which exist in two configurations, α and β , with corresponding energies ε_{α} and ε_{β} , and lengths a and b. The chain is subject to a constant tensile force f.



- (a) Write the partition function Z_N for the system. (Account for the potential energy associated with f in addition to the individual energies ε_{α} and ε_{β} . You could think of the work involved in stretching each molecule to its current length.)
- (b) Calculate the average length $\langle L \rangle$ as a function of f and the temperature T.
- (c) Assume that $\varepsilon_{\alpha} > \varepsilon_{\beta}$ and a > b. Give the average length $\langle L \rangle$ in the absence of the tensile force f = 0. What are the high- and low-temperature limits, and what is the characteristic temperature at which the changeover between the two limits occurs?